

SAB Draft Report to Assist Meeting Deliberations -- Do not Cite or Quote -- This draft is a work in progress, has not been reviewed or approved by the chartered SAB, and does not represent EPA policy.

1
2 UNITED STATES ENVIRONMENTAL PROTECTION AGENCY
3

4
5 OFFICE OF THE ADMINISTRATOR
6 SCIENCE ADVISORY BOARD
7

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10 EPA-SAB-07-XXX

11
12 The Honorable Stephen L. Johnson
13 Administrator
14 U.S. Environmental Protection Agency
15 1200 Pennsylvania Avenue, NW
16 Washington, DC 20460

17
18 Subject: Science Advisory Board (SAB) Report on Office of Pollution
19 Prevention and Toxics (OPPT) Estimation Programs Interface
20 Suite (EPI Suite™)
21

22 Dear Administrator Johnson:
23

24 At the request of the Office of Pollution Prevention and Toxics (OPPT), the SAB
25 recently reviewed the OPPT's Estimation Programs Interface Suite (EPI) Suite™
26 software. The Agency uses EPI Suite™ to support regulatory decisions in the new
27 chemicals program and in other chemical assessment activities.
28 .

29 The SAB commends EPA for the strategic decision to develop EPI Suite™ and to
30 make it easily and freely available. Governmental and private organizations within the
31 United States and elsewhere make extensive use of the software in supporting decisions
32 regarding new and existing chemicals. The multiplicative and widespread uses of EPI
33 Suite™ stems, in part, from its successful utilization and integration of available science
34 in combination with its user-friendliness, transparency, and cost-effectiveness. Because
35 EPI Suite™ is part of the Organization for Economic Co-operation and Development's
36 quantitative structure-activity relationship ((Q)SAR) toolbox, for example, the software
37 will likely play a significant role in the implementation of the European Union's
38 Registration, Evaluation and Authorization Chemical Policy. EPI Suite™ also supports
39 emerging industrial economies to develop in an environmentally protective and
40 sustainable manner.

41
42 The EPI Suite Review Panel has carefully evaluated the EPI Suite™ software.
43 The Panel's numerous recommendations for improvements in the software's scope,
44 accuracy, and ease of operations appear in the enclosed report, along with comments on
45 appropriate current and potential future uses. Because of its importance in supporting
46 Agency decisions regarding existing and new chemicals, the Panel would like to draw
47 your attention to the following two overarching findings.

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1
2 First, for chemicals similar to those for which chemical property estimating
3 modules were developed, the EPI Suite™ calculations are sufficient to support Agency
4 regulatory screening applications. However, for existing and/or new chemicals whose
5 structures and/or properties are outside the domain used in module development,
6 uncertainty considerations may limit the utility of EPI Suite™ results. In such cases, the
7 Agency uses other methodologies to evaluate chemicals.

8
9 Secondly, the resources for maintaining and improving EPI Suite™ have not been
10 commensurate with its importance in supporting Agency decisions nor with the rapidity
11 with which new and even novel chemicals are being developed for commercial use. In
12 light of its widespread and multiple uses, the Agency should increase its efforts to
13 expand, to the extent possible, the range of chemical categories over which the software
14 can generate valid predictions and to expand the suite of modeled chemical properties as
15 new scientific information becomes available.

16
17 Thank you for the opportunity to provide advice on this important suite of
18 modeling software and to interact with the very dedicated and able OPPT staff. Please
19 feel free to contact us if you have any questions concerning this review.

20
21
22
23 Sincerely,

24
25
26
27
28 Dr. Granger Morgan, Chair
29 EPA Science Advisory Board

Dr. Michael J. McFarland, Chair
EPI Suite Review Panel
EPA Science Advisory Board

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NOTICE

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This report has been written as part of the activities of the EPA Science Advisory Board, a public advisory group providing extramural scientific information and advice to the Administrator and other officials of the Environmental Protection Agency. The Board is structured to provide balanced, expert assessment of scientific matters related to the problems facing the Agency. This report has not been reviewed for approval by the Agency and, hence, the contents of this report do not necessarily represent the views and policies of the Environmental Protection Agency, nor of other agencies in the Executive Branch of the Federal government, nor does mention of trade names or commercial products constitute a recommendation for use. Reports of the EPA Science Advisory Board are posted on the EPA website at <http://www.epa.gov/sab>.

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2 **Science Advisory Board**
3 **EPI Suite Review Panel**
4

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EXECUTIVE SUMMARY

1
2
3 The Office of Pollution Prevention and Toxic Substances (OPPT) is responsible
4 for assuring the public that industrial chemicals for sale and use in the United States do
5 not pose unacceptable risks to human health or the environment. To accomplish this,
6 OPPT promotes pollution prevention, use of safer chemicals, risk reduction, risk
7 management and public awareness. OPPT programs include the pre-manufacture
8 notification (PMN) review of new industrial chemicals; testing, assessment, and risk
9 reduction of existing industrial chemicals; management of “national chemicals” (e.g.
10 PCBs); international chemical issues; pollution prevention advocacy; and partnership
11 programs, such as the High Production Volume Chemicals (HPV) Challenge, Green
12 Suppliers Network, Design for the Environment and Green Chemistry.

13
14 Accurate and reliable predictions of the behavior of chemicals in a biological or
15 environmental system require a full and comprehensive understanding of their
16 thermodynamic, kinetic and transport properties both within and across multimedia
17 compartments. To support Agency decisions regarding the toxicity, environmental fate
18 and transport of new chemicals, OPPT (with Syracuse Research Corporation (SRI))
19 developed the Estimation Programs Interface (EPI Suite™), which OPPT makes freely
20 available from its website. The software combines the available science with user-
21 friendliness, transparency, and cost-effectiveness. EPI Suite™ is utilized by various
22 Agency program offices as well as other US federal agencies, state regulatory agencies,
23 foreign countries and the private sector.

24
25 The EPI Suite™ software consists of physical-chemical property estimation
26 routines (PERs) and mass balance based environmental fate models (EFMs). Where
27 measured data are lacking and EPI Suite™ is appropriate, the Agency uses the results of
28 the PERs together with the EFMs, to understand a chemical’s environmental fate and
29 transport. This understanding is fundamental to assessing chemical exposure, hazard,
30 and risk.

31
32 OPPT requested that the Science Advisory Board (SAB) evaluate the science,
33 functionality and uses of the Agency’s EPI Suite™ software. The EPI Suite Review
34 Panel was formed for this purpose and reviewed the software in the context of OPPT’s
35 needs.

36
37 **Science.** In summary, the Panel commends the Agency for using sound science
38 to develop and refine EPI Suite™ and encourages the further development and use of this
39 software in supporting Agency decisions. The Panel applauds the Agency for furnishing
40 chemical fate and transport modeling software that is science-based and is used globally
41 to support environmental policy decisions.

42
43 The Panel encourages the Agency to consider evaluating the chemical fate and
44 transport modules using the latest statistical approaches to determine their predictive
45 accuracy and to evaluate new estimation approaches as they gain acceptance in the

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1 scientific community. The Panel endorses a systematic approach for updating and
2 refining the chemical fate and transport modules as high quality and peer-reviewed
3 measurement data become available – both to increase the applicability of the software to
4 a wider array of chemical classes and to support the inclusion of additional physical-
5 chemical properties. The Panel has provided a number of recommendations focused on
6 expanding the current set of chemical properties and associated functionality including
7 EFM's for future upgrades to EPI Suite™. However, in light of the widespread
8 application of EPI Suite™, the Panel recommends that before the Agency decides to add
9 a module, it assess, to the extent practical, whether there is consensus in the scientific
10 community that the module has been appropriately parameterized and has been
11 sufficiently verified to be applicable in screening assessment. Also, because the
12 accuracy of EPI Suite™ output will vary depending on the chemical and the
13 environmental compartment in which it is found, the Panel recommends communicating
14 the uncertainty associated with estimates provided by EPI Suite™.

15
16 The PERs currently within EPI Suite™ have received extensive scientific
17 scrutiny with the results published in the peer-reviewed literature. Because EPI Suite™
18 was historically developed to model the fate and transport behavior of nonpolar organic
19 chemicals, the physical-chemical property estimates for this class of chemicals are
20 typically well within an order of magnitude of measured values. The Panel considered
21 these results adequate to support Agency screening level decision-making. Moreover,
22 these PERs satisfy the Organization for Economic Cooperation and Development
23 (OECD) principles established for quantitative structure-activity relationship ((Q)SAR)
24 validation, a finding which further supports the use of EPI Suite™ PERs in screening
25 level regulatory decision-making.

26
27 The ability of EPI Suite™ to accurately model physical-chemical properties
28 depends on the chemical's class, the quality of the property module chemical data
29 training set and whether the chemical's properties fall within the range of the chemical
30 training data set. Many of the chemical training data sets are outdated and some are
31 incomplete. Periodic review and refinement of the training sets would support the
32 continuous improvement of module output accuracy and expand the range over which
33 EPI Suite™ results are valid. These refinements could be accelerated if the Agency
34 leveraged its resources to collect additional measured property data. Criteria that the
35 Agency should consider in prioritizing the updates of chemical property data sets are
36 identified in 1-A-ii below.

37
38 Chemical domain mapping has the potential to significantly improve the
39 predictive capabilities of mechanistically and statistically-based PERs, but no Panel
40 consensus emerged as to the most effective approach to achieve this goal. The Panel
41 encourages the Agency to consider establishing a scientific forum at which the various
42 methodologies for enhancing the accuracy of the PER module output may be evaluated.

43
44 The Panel agreed on two broad recommendations aimed at improving EFM
45 module predictions. First, the Panel supports a more explicit description and justification
46 for the Agency's selection of EFM parameter default values. Secondly, the Panel

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1 encourages the Agency to provide the EPI Suite™ user with a clear and unambiguous
2 display of quantitative uncertainty estimates associated with the fate model (i.e., EFM)
3 output.
4
5
6

7 **Functionality.** The Panel, which included experienced as well as novice users of
8 EPI Suite™, considered the functionality and usability of EPI Suite™ software. While
9 there are many positive features associated with the EPI Suite™ user interface including
10 its documentation and HELP file availability, there are also opportunities for functional
11 improvement. For example, although EPI Suite™ operates within a Windows™
12 platform, a new user to EPI Suite™ is immediately struck by the disk operating system
13 (DOS) appearance of the graphical user interface (GUI). The Panel encourages the
14 Agency to secure the necessary funding to upgrade EPI Suite™'s GUI to reflect a typical
15 Windows™ appearance and functionality.
16
17

18 **Uses.** All of the modules in EPI Suite™ are generally accepted by the regulatory
19 and regulated community for use in risk-based priority setting, screening level risk
20 assessment and prioritization for chemical testing for the chemical classes to which the
21 modules apply. Given the mandated 90-day reporting period for which new chemicals in
22 the PMN program must be evaluated and the large number of chemicals that the Agency
23 must screen annually, reliance on (Q)SAR module output is justified. The modules are
24 expected to provide an order of magnitude estimate of a chemical's physical properties,
25 an accuracy level that is generally acceptable by the Agency for screening level
26 assessments. However, application of (Q)SAR-based modules to chemicals outside the
27 module training set domain increases the uncertainty of the module prediction. Because
28 the chemical domains that are used in developing current EPI Suite™ (Q)SARs do not
29 provide adequate coverage of nanoparticles, inorganic compounds, organo-metallic and
30 certain other classes of chemicals, application of EPI Suite™ for these classes of
31 compounds within the PMN and pollution prevention (P2) programs is inappropriate.
32 The Panel recommends that the Agency collect more peer-reviewed measurement data on
33 the physical and chemical properties for these chemical classes with the intent of either
34 expanding the domain of the existing (Q)SARs or for creating new (Q)SARs specifically
35 for these classes of chemicals.
36

37 Owing to its success in supporting Agency decision-making and its accessibility,
38 use of EPI Suite™ is prolific outside of the Agency, including in international regulatory
39 agencies. Given its broad acceptance and use by regulators, industry and the academic
40 community, the Panel strongly encourages the Agency to explore opportunities to
41 develop foreign language versions of EPI Suite™.
42
43
44
45

RESPONSE TO THE CHARGE

Before responding directly to the Agency charge questions, the Panel thought it appropriate to furnish the broader scientific audience with a brief description of the organizational structure, technical underpinnings and Agency uses of the EPI Suite™ modules. The intent of this section of the report is to establish a scientific and regulatory context with which to evaluate the Panel's findings and recommendations.

The EPI Suite™ software basically consists of two module categories: physical-chemical property estimation routines (PERs) and environmental fate models (EFMs). The PERs are used to predict important physical-chemical (e.g. water solubility, vapor pressure, octanol-water partition coefficients) and reactivity (e.g. biodegradation, atmospheric oxidation) properties and, together with the EFMs, project a chemical's environmental fate and transport which is considered during the Agency's screening level evaluation.

Accurate and reliable predictions of the behavior of chemicals in a biological or environmental system require a full and comprehensive understanding of their thermodynamic, kinetic and transport properties both within and across multimedia compartments. To support Agency decisions regarding the toxicity, environmental fate and transport of new chemicals, the EPI Suite™ software employs twelve individual modules that may be logically placed into one of these two functional categories.

Category – 1: The nine regression based estimation modules in the PER category were developed for estimating physical-chemical properties for chemicals that lack the minimum data set needed to support Agency decisions. These modules, including the Octanol-Water Partitioning Coefficient Estimation Program (KOWWIN), the Henry's Law Constant Estimation Program (HENRYWIN), the Soil or Sediment Organic Carbon Partitioning Coefficient Estimation Program (PCKOCWIN), the Water Solubility Estimation Program (WSKOWIN), the Biocentration Factor Estimation Program (BCFWIN) and the Melting Point-Boiling Point (and Vapor Pressure) Chemical Estimation Program (MPBPWIN), MPBPPVWIN), are used for estimating the equilibrium distribution or partitioning of a chemical between two media such as fish tissue-water and organic matter-water (which are functions of the octanol-water partition coefficient), air-water, organic matter-water, etc. The three other modules found in the PER category include: the Atmospheric Oxidation Estimation Program (AOPWIN), the Biodegradation Estimation Program (BIOWIN) and the Hydrolysis Estimation Program (HYDROWIN). These modules employ regression-based approximation methods to estimate the value of kinetic parameters for atmospheric gas-phase reaction with the hydroxyl, aerobic biodegradation and hydrolysis reactions, respectively.

Category - 2: EPI Suite™ EFM modules that enable the user to estimate the environmental fate and transport of specific chemicals include: the Volatilization Rate from Water Estimation Program (WVOLWIN), the Sewage Treatment Plant Chemical

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1 Fate Estimation Program (STPWIN) and multi-media fugacity model (LEV3EPI). These
2 modules, which utilize the Lavoisier chemical species mass balance approach, have been
3 designed to estimate the chemical concentration, phase mass fractions and residence
4 times of chemicals when placed in well-defined environmental systems. The mass
5 balance approach allows the user to estimate the change in chemical concentration over
6 time from which removal rates can be estimated. Moreover, the EFM modules employ,
7 as inputs, the partitioning and reaction kinetic results generated from the PER modules.
8 The EPI Suite™ user, however, has the ability to override these default inputs and enter
9 their own values.

10
11 The environmental compartments defined within the three EPI Suite™ EFM
12 modules require the user to input the volume and mass fractions of the various media
13 under consideration. In the absence of user defined values, EPI Suite™ assigns default
14 values, which are idealized representations of the real world. Requirements of the EFM
15 modules also include user (or EPI Suite™ – i.e., default) defined chemical coefficients
16 that quantitatively describe the rate of chemical transport between the various media
17 compartments.

18
19 An important limitation of the present version of EPI Suite™ is the inability for
20 users to input their own mass transfer coefficient (MTC) data. Moreover, the absence of
21 high quality peer-reviewed MTC data to serve as input to EPI Suite™ exacerbates this
22 problem. Although filling this critical data gap is vital for broadening the range of
23 applicability of EPI Suite™, collecting useful MTC data is inherently expensive, a fact
24 which presents the Agency with a considerable resource challenge. Because of the
25 importance of obtaining and incorporating accurate and reliable MTC information into
26 EPI Suite™, the Panel encourages the Agency to develop a systematic and longer-term
27 program, possibly through leveraging resources with other federal agencies, to address
28 this critical data need. However, in the interim, the Panel endorses establishing a modest
29 effort that can, at a minimum, result in the formulation of a guideline MTC module based
30 on available peer reviewed theoretical models and supporting data. A workshop
31 consisting of an expert panel sponsored by the Agency is suggested as a means of
32 producing a draft of the guideline version of the MTC module. A summary assessment
33 of core EPI Suite™ modules can be found in APPENDIX 2.

34
35 Two related, but separate, models were not addressed in this review. These were
36 the Ecological Structure Activity Relationship Program (ECOSAR), which comes
37 “bundled” with EPI Suite™ and the Dermal Permeability Coefficient Program
38 (DERMWIN), which is included in the EPA Zip file containing EPI Suite™.

39
40 The remainder of this report is organized according to the charge provided by the
41 program office.

1 **1. Supporting Science**

2 **A. Comprehensiveness**

3 **i. Are there additional properties that should be included in upgrades**
4 **to EPI Suite™ for its various specified uses (PMN, P2)?**

5
6 All of the physical-chemical properties that are currently modeled by EPI Suite™
7 are critical in characterizing the behavior of a chemical released into the environment.
8 Therefore, none should be dropped.

9
10 Under most circumstances, the PERs predict the measured property value within
11 an order of magnitude, a standard of accuracy that is generally acceptable for screening
12 level Agency decision-making. It would be inappropriate to use PERs to predict
13 physical-chemical properties of chemicals whose characteristics are significantly
14 different than those found in the module training set because the difference between
15 predicted and measured values may be greater. This potential inaccuracy is an important
16 issue unto itself and also for error propagation when these estimates are incorporated into
17 the fate models.

18
19 Given the broad range of chemicals for which the Agency must prepare
20 environmental assessments together with the need to ensure an equitable and transparent
21 evaluation of all chemical data submissions, the Panel encourages the Agency to furnish
22 stakeholders with a description of the process by which regulatory decisions are made for
23 chemicals when application of EPI Suite has been determined to be inappropriate.

24
25 With respect to expanding the current set of chemical properties (and associated
26 functionality) for future upgrades to EPI Suite™, the Panel recommends that the Agency
27 consider incorporating the following:

- 28
- 29 • pK_a , the negative log of a chemical's dissociation constant
 - 30 • Influence of pK_a on other physical-chemical properties
 - 31 • Temperature dependency of all physical-chemical properties
 - 32 • K_{AW} , the air-water partition coefficient
 - 33 • K_{OA} , the octanol-air partition coefficient
 - 34 • Bioaccumulation factors for root plants, leaf plants, and aquatic wildlife
 - 35 • Diffusion coefficients in various environmental media
 - 36 • Metabolism and production of stable chemical intermediates
 - 37 • Neutral hydrolysis
 - 38 • Activity coefficients
 - 39 • Sub-cooled liquid vapor pressure and aqueous solubility
 - 40 • Surface tension
 - 41 • Anaerobic biodegradation potential
 - 42 • Ozone depletion potential, greenhouse gas potential, and maximum incremental
43 reactivity (MIR) used to evaluate ozone formation potential.
- 44

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1 Some of these endpoints and improved features (e.g. temperature-dependence of
2 physical-chemical properties) can already be predicted by another Agency supported
3 model (SPARC). The Panel, therefore, encourages the Agency to consolidate and build
4 upon existing work for future EPI Suite™ improvements.

5
6 The current EPI Suite™ has only limited utility in predicting parameters for the
7 important and large class of compounds known as polymers. Several Panel members
8 offered the following list of additional chemical properties specifically related to the
9 toxicity and fate of polymers that the Agency may consider in future upgrades to EPI
10 Suite™:

- 11
- 12 • Glass transition temperature
- 13 • Crystal melt transition temperature
- 14 • Elastic mechanical properties like bulk modulus
- 15 • Viscosity measures
- 16 • Heat capacity
- 17 • Cohesive energy
- 18 • Charge
- 19 • Water solubility
- 20 • Dispersibility
- 21 • Flammability
- 22 • Parameters (e.g., degradation rates) influencing environmental persistence
- 23

24 Several commercial software packages estimate many of the environmentally
25 important physical-chemical properties of polymers. The Panel encourages the Agency
26 to evaluate the scientific underpinnings of these software packages to determine if similar
27 functionality could be incorporated into EPI Suite™.

28
29 For some classes of chemicals, the physical-chemical properties estimated by EPI
30 Suite™ are not sufficient to predict a chemical's behavior. The Panel encourages the
31 Agency to consider development of a systematic and longer-term plan to develop and
32 integrate additional EPI Suite™ functionality to adequately model additional physical-
33 chemical properties as well as the fate and transport characteristics of these compounds.
34 Similarly, the Panel strongly recommends that the Agency establish and support technical
35 transfer symposia and associated activities (e.g., science workshops) that will help
36 facilitate Agency exposure to the latest scientific approaches to chemical property
37 modeling.

38
39 Given the Agency's resource limitations, the Panel strongly recommends that the
40 Agency establish a set of objective and transparent criteria for identifying and prioritizing
41 the most important physical-chemical properties required for defensible regulatory
42 decision-making. Examples of possible ranking criteria, which are not listed in any sort
43 of priority, include the following:

- 44
- 45 • The property's potential use in future fate and transport modeling enhancements
- 46

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- 1 • The accuracy and reliability of the property's currently available experimental
2 data set
- 3
- 4 • The extent of the chemical domain covered by the modeled property
- 5
- 6 • The opportunity for increasing the scope and applicability of EPI Suite™ to a
7 broader range of chemical classes and properties.
- 8
- 9 • Determination of whether the new property could be easily modeled using the
10 existing model chemical data set
- 11
- 12 • Relative importance of property value as input to other EPI Suite™ modules
13 and/or Agency chemical assessments
- 14
- 15 • The relative magnitude between "model error" and "measurement error"
- 16
- 17 • Cost or other resource requirements associated with modeling the new property
- 18
- 19

20 Greater use of MTCs can improve some applications in EPI Suite™. A recent
21 study comparing the outputs of five multimedia models demonstrated that model
22 homogenization was possible only when the numerical values of the dozen or so MTCs
23 were numerically equal (Cowan, et al., 1995). Where MTCs varied significantly, the
24 computed concentration levels, mass fractions in the media compartments and the
25 chemical residence time estimates differed, in many cases, by several orders of
26 magnitude. The peer-reviewed literature contains a significant quantity of data with
27 which to develop MTCs. Therefore, the Panel encourages the Agency to support the
28 development of additional MTCs and, where possible, establish a systematic process for
29 evaluating and incorporating high quality MTC data within EPI Suite™

30
31 The highest priority fate models are those which are judged to be used most often
32 and/or to have the most impact on decision-making processes. The Panel has identified
33 these models to be:

- 34
- 35 • Fugacity Unit World
- 36 • STP
- 37 • BCF/BAF
- 38 • Long-range transport
- 39

40 While there was consensus among the panelists that BAF is an important fate
41 parameter to model and the Panel encourages the EPA to develop this module, several
42 panelists strongly cautioned that BCF/BAF models still have an incomplete treatment of
43 certain factors important in predicting uptake and metabolism. For example, while the
44 Arnot and Gobas (2004) model includes a metabolism term, it is not clear, given
45 experimental difficulties, how accurately this term can be parameterized for different
46 compounds in different biota. For metabolizable chemicals (e.g. aliphatic alcohols or

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1 acids that have predicted log Kow values greater than 5 but are readily metabolized), the
2 predictions of BCF and BAF from a model based solely on log Kow can be significantly
3 greater (e.g., one order of magnitude or more) than experimentally determined BCF
4 values. While this type of phenomenon has been recognized by researchers involved in
5 development of BCFWIN (Meylan et al., 1999), and since the module in EPI Suite™
6 does contain correction factors to attempt to account for metabolism, further work is
7 needed to improve its predictive capability.

8
9 Some panelists identified related concerns with the development of this module,
10 including:

- 11
12 • Conducting experimental studies for BAF to validate the model is difficult and
13 expensive and such studies have been conducted only for a limited number of
14 substances which are either slowly or not metabolized.
- 15
16 • Within the literature there are wide ranges reported in field measured BAFs (and
17 even BCFs in laboratory studies) that have been obtained for a given chemical.
- 18
19 • Concern was expressed regarding the difficulty in appropriately parameterizing a
20 BAF model for non-recalcitrant chemicals. A correction factor approach alone (as
21 is used in BCFWIN) may still lead to significant errors in prediction for certain
22 substances (or potential errors where measurement data are not available), and
23 novice users may not appreciate the limitations in these predictions.
- 24
25 • There is no widely accepted method for estimating whole body metabolism rates
26 in fish either from first principles (i.e., structure or other properties) or otherwise
27 although there is considerable research on-going to develop and validate such
28 methods. These efforts include the International Life Sciences Institute/Health
29 and Environmental Sciences Institute (ILSI/HESI) project and recently initiated
30 work by ECVAM. Therefore, even if the user were given the option to enter a
31 metabolism rate, these estimates are not currently available.
- 32
33 • There is the potential for inconsistencies between the outputs of BCFWIN and the
34 potential new BAF modules (e.g., Arnot and Gobas model) that may lead to
35 confusion in the interpretation of the fate of some chemicals in part because these
36 two models are based on very different approaches. BCFWIN relies on a fitted
37 equation to measured BCF data. The Arnot and Gobas model is based on first
38 principles and, as such, includes hydrophobic partitioning, growth dilution, and
39 metabolism. When differences between the model predictions represent the
40 variability in BCF and BAF data this is acceptable. However, in many cases, the
41 differences will be due to problems in adequately parameterizing the BAF model
42 (e.g., to account for metabolism) and it would be difficult to know that this is the
43 cause of the discrepancy *a priori*.
- 44

45 The training set used to calibrate the existing model, BCFWIN, includes studies
46 based on analysis of parent test substance as well as studies based on analysis of total

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1 radioactivity. The total radioactivity based BCF can not distinguish between parent
2 substance bioaccumulation and incorporation of metabolites into the organism as a result
3 of normal catabolic processes (although the Panel recognizes that some metabolites can
4 be of toxicological concern). As a result, the model is trained on data that lacks a
5 consistent basis for (Q)SAR development and subsequent decision-making. The
6 BCFWIN database also fails to indicate whether the basis for the BCF is parent substance
7 or total radionuclide analysis.

8
9 Given the increasing focus on the assessment of persistent, bioaccumulative and
10 toxic (PBT) chemicals in regulatory contexts, the current BCFWIN data set should be
11 critically reviewed, any inappropriate data that does not meet acceptance standards (e.g.,
12 total radioactivity based BCF for metabolized substances) deleted, and new literature data
13 added to provide a consistent basis for an improved "next generation" (Q)SAR.

14
15 The existing Japanese "MITI" BCF database provides perhaps the best best single source
16 of aqueous fish BCF data that could be included in this effort. The data in the MITI
17 database is based on the OECD 305 bioaccumulation test procedure, which is currently
18 considered by many to be the "gold standard" for these types of tests
19 (http://www.safe.nite.go.jp/english/kizon/KIZON_start_hazkizon.html). Compilation of
20 such data also would support the development of (Q)SARs for estimating fish
21 biotransformation potential that could be used as input to BAF models or multimedia
22 exposure models that predict human intake fraction.

23
24 The panelists encourage the Agency to participate in and follow the on-going
25 scientific developments in BAF determinations including:

- 26
27 • Additional efforts at experimentally determining bioaccumulation (including
28 better understanding metabolism)
- 29
30 • Improved databases for developing and verifying BAF models
- 31
32 • ILSI/HESI (International Life Sciences Institute/Health and Environmental
33 Sciences Institute) Work Group on Bioaccumulation
- 34
35 • Ongoing modeling research published in the literature

36
37 In light of the widespread application of EPI SuiteTM, before the decision is made
38 to add a new module, such as the BAF module, the Agency should assess to the extent
39 practical, whether there is consensus in the scientific community that the model has been
40 or can be appropriately parameterized and has been sufficiently verified to be applicable
41 in screening assessments.

42
43 Related issues are discussed in section 1-C-ii below. More detailed information
44 can be found in the APPENDIX for 1Ai.

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1 **ii. Are there additional sets of existing measured data which should be**
2 **included in upgrades to EPI Suite™? Are there specific measurements**
3 **with the potential to improve EPI Suite™ estimates so much that an effort**
4 **should be made to collect them?**

5
6 Existing peer-reviewed measurement data sets are available for the following
7 parameters: octanol-water partition coefficients (K_{ow}), Henry's law constants (H_C), air-
8 octanol partition coefficients (K_{Ao}), biodegradation rates, organic carbon partition
9 coefficient (K_{oc}), aqueous solubility, and rates of aquatic hydrolysis. Several panelists
10 noted that updating the chemical training data set used in estimating K_{oc} should be a
11 priority because of the limited amount of data that is currently used to estimate the value
12 of this parameter within EPI Suite™. The Panel encourages the Agency to expand the
13 functionality of the K_{oc} module to capture the range of organic carbon types that could
14 affect a chemical's fate and transport including: natural vegetation-based, soot, black
15 carbons, non-aqueous phase liquids (NAPL), etc. Appendix 1-B-i identifies additional
16 data sets the Agency might consider.

17
18 Because of the Agency's limited resources, the Panel supports a strategic
19 approach to identifying those data sets that require refinement. Criteria that the Agency
20 should consider in prioritizing the updates of chemical property data sets include the
21 following:

- 22
23 • The duration of time since the chemical property data set was last updated
24 • Level of uncertainty associated with the chemical property estimates
25 • The domain and quality of the chemical property training set domain
26 • Accuracy of chemical property prediction

27 Several panelists identified scientific proceedings associated with certain highly
28 reputable international conferences and journals such as the J. Phys. Chem. Ref. Data
29 (<http://jpcrd.aip.org>) as excellent sources of peer reviewed chemical data sets that should
30 be considered for inclusion in upgrades to EPI Suite™. There are additional sets of
31 measured data that the Agency could consider for inclusion in upgrades to EPI Suite™
32 pending the Agency's satisfaction with the quality of peer-review received. Some of
33 these are:

- 34
35 • Additional sewage treatment plant (STP) chemical partitioning and fate data.
36 Appropriate sources for this type of data would include, but are not limited to: a)
37 the the National Association of Clean Water Agencies (formerly the Association
38 of Metropolitan Sewerage Agencies), b) Water Environment Research Foundation
39 (WERF), c) Water Environment Federation (WEF) and d) Journal of
40 Environmental Engineering and related journals.
41
42 • The existing Japanese "MITI" data - While most data bases aggregate data from a
43 number of different studies using different methods, the MITI database uses a
44 standard procedure to test a large number of chemicals, including direct

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1 measurement of the properties of interest for parent compounds. Some panelists
2 familiar with the database say it provides an excellent source of aqueous fish BCF
3 data.

- 4
- 5 • Additional sources of Polychlorinated Biphenyls (PCB) congener data sets that
6 are available in the peer-reviewed literature (e.g., Frame et al., 1996a, 1996b)
7
 - 8 • Reliable un-published data reported as part of the High Production Volume
9 (HPV) challenge program (<http://www.epa.gov/HPV/>) or other international
10 regulatory initiatives such as the OECD Screening Information Data Set (SIDS)
11 program (<http://www.epa.gov/opptintr/chemtest/pubs/oecdsids.htm>).
12

13 The Panel agreed that the EPI Suite™ fate and transport modules are limited by the
14 paucity of chemical degradation (e.g. biodegradation and biotransformation processes)
15 data available. Like mass transfer coefficients, chemical degradation information is so
16 important to understanding the fate and transport of chemicals in the environment that, if
17 necessary, the Agency should consider redirecting resources from current programs to
18 address this critical data need. Moreover, there have been a number of recent scientific
19 advances in understanding chemical degradation that merit Agency consideration. For
20 example, an innovative methodology termed the environmental “reagents” approach has
21 been developed for defining the reactive power of environmental compartments.
22 Understanding this reactivity has important implications to the fate of chemicals and
23 should be considered in future upgrades to the EPI Suite™ chemical degradation modules
24 (Green and Bergman 2005).
25

26 **iii. Are there other capabilities that should be included in upgrades to**
27 **EPI Suite™? The Agency is especially interested in the SAB’s views on**
28 **uncertainty analysis and if/how information on how good the estimates are**
29 **can be conveyed to users.**

30

31 **Uncertainty in Parameter Estimation, Routines, and Predictions**

32

33 When a PER is used to predict properties for chemicals lying outside the domain
34 of compounds used in the training set for that PER, confidence in the prediction will
35 generally be lower than if the chemical were within the existing domain. The Panel
36 recommends that results in such cases be flagged to highlight for the user the potential
37 uncertainties in the estimate value.

38

39 Although the Panel explored a range of views concerning how uncertainty should
40 be conveyed to the EPI Suite™ user, two approaches emerged as the preferred options.
41 Both approaches involve the development of appropriate statistical confidence intervals
42 surrounding a mean value of an estimated chemical property. In the first case, the
43 majority of the Panel recommended that the quantitative uncertainty information be
44 displayed only in HELP files while, in the other, several panel members preferred having

1 the data presented with the module output for each endpoint/test chemical. Advantages
2 and disadvantages of both approaches are summarized in the following:
3

- 4 • Provide information on the confidence range in HELP files:
5

6 Advantage: This approach does not require that the Agency defend quantitative
7 estimates, particularly for test chemicals that are outside of the model domain.
8 Moreover, by limiting the availability of the uncertainty discussion to the help
9 file, the Agency reduces the potential for misinterpretation or misapplication of
10 the uncertainty results.

11
12 Disadvantage: If not presented more explicitly, the novice user may overlook
13 this information increasing the potential for misinterpretation or misapplication
14 of the model results.

- 15
16
17 • Provide the confidence interval in the module output:
18

19 Advantage: The Agency and the scientific community are moving toward
20 more explicit acknowledgement and quantification of uncertainty. This
21 approach is consistent with such goals. Moreover, by including quantitative
22 uncertainty estimates with module output, the EPI Suite™ user is compelled to
23 recognize the potential of making decision errors.
24

25 Disadvantage: While the complex nature of data uncertainties and modeling
26 uncertainties needs to be communicated, more informative, but potentially
27 more complex, quantitative uncertainty assessment methods present novice
28 users and decision makers with new challenges. Effective incorporation of
29 uncertainty in decisions will not be accomplished with quantitative uncertainty
30 analysis alone.
31

32 The Panel encourages the Agency to explicitly acknowledge to the EPI Suite™
33 user the fact that the quantitative uncertainty estimate for each endpoint/test chemical
34 includes only the statistical error associated with the model prediction and neglects the
35 error in reported experimental measurement values that were used to calibrate the model.
36 To the extent practical, the Agency should provide guidance to the user on the expected
37 data error component for each modeled property.
38
39

40 **Uncertainty in Environmental Fate Model Predictions**

41

42 The Panel endorses that uncertainty associated with the EPI Suite™ fate model
43 (i.e., EFM) be better conveyed to the user. The Panel identified the following sources of
44 EFM uncertainty:
45

- 46 • Model structure

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- 1 • Model parameters (e.g., chemical properties, mass transfer coefficients, etc.)
- 2 • Media compartment (s) including type, size and distribution

3
4 Panel deliberations included consideration of various approaches to effectively
5 convey uncertainty to the EPI Suite™ user. The following list summarizes the range of
6 approaches discussed by the Panel together with their potential advantages and
7 disadvantages.

- 8
9 • Model output details could remain in its current form, while the documentation
10 could more fully describe the input parameter range and limitations of the
11 evaluative fate models.

12
13 The EFM modules in EPI Suite™ are designed to produce “evaluative”
14 predictions. The media compartments reflect generic environmental scenarios such as the
15 “unit world”. The term evaluative is used to describe an output that is interpreted to be of
16 relative significance and/or order-of-magnitude rather than a precise numerical result.
17 The major (i.e. 1st order) sources of output uncertainty are associated with the ascribed
18 media of chemical entry. For example, significantly different media concentration
19 predictions will result if the chemical is “emitted” into the air compartment rather than
20 the water compartment. Clear data/information available in the PMN as to the choice of
21 media for chemical entry is needed. In addition, cautions/alerts as to the high level of
22 output variability resulting from media entry choice need be placed in the documentation
23 as understanding this variability is key to controlling this source of EFM output
24 uncertainty. Experience with such models indicates that input variations in chemical
25 properties and MTCs result in 2nd order levels of EFM output uncertainty (Webster, et.
26 alAl, 1998)

27
28 Advantages: Simplicity and consistency in interpretation of fate model output.

29
30 Disadvantages: Only presenting uncertainty information in the help section
31 assumes that the user will read this section. Even if this section were read,
32 there is no guarantee that the scientific or regulatory implications of
33 uncertainty will be fully understood.

- 34
35 • Give qualitative information regarding the uncertainty associated with model
36 results based on the range of the chemical property values.

37
38 An example of such an approach is illustrated by describing a chemical’s
39 distribution using a K_{OA} versus K_{OW} diagram. Construction of such a plot will depict the
40 distribution of the chemical with respect to the various environmental phases, e.g., air,
41 water or soil/sediment. EPI Suite™ should provide explanatory text that clearly informs
42 the user that the relative media compartment sizes, inter-compartmental chemical mass
43 transfer rates and the media compartment into which the chemical is released will affect
44 the model predictions of the chemical’s allocation between media compartments.
45 Moreover, if a chemical were associated exclusively with a single medium, uncertainty in

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1 the partition coefficients would have a minimal impact on the chemical's allocation
2 between compartments (as compared to those chemicals that are distributed between
3 phases).

4
5 Advantages: The user will receive qualitative information regarding the
6 potential sensitivity of model output to physical-chemical properties as it
7 relates to environmental fate. This approach provides yet another level of
8 screening whereby a chemical that does not clearly lie exclusively within a
9 specific environmental compartment may merit further investigation (based on
10 environmental partitioning concerns alone).

11
12 Disadvantages: Development of a robust method for determining and
13 presenting this information represents a considerable technical challenge.

- 14
15 • Calculate error propagated from estimates of physical-chemical properties and
16 fate models, i.e., input 95% confidence limits or qualitative confidence factors
17 from each estimated physical-chemical property to obtain a range of fate results
18 (MacLeod et al. 2002).

19
20 MacLeod et al. (2002) present a simple, semi-quantitative method for calculating
21 error propagated through environmental fate models. Several panel members supported
22 this approach over the computationally demanding Monte Carlo simulation where the
23 required number of model iterations can be significant (e.g., > 2000 iterations). The
24 semi-quantitative approach provides a simple view of the range of values that could be
25 expected based on user-defined uncertainties associated input parameters where
26 uncertainty is expressed as a multiplicative factor.

27
28 Advantages: With this method, the user generates an estimate of the
29 distribution of the model output for each chemical in the various media
30 compartments. Use of this approach assumes that the user will have an
31 estimation of the uncertainty associated with the model inputs.

32
33 Disadvantages: The uncertainty associated with other factors (e.g., mass
34 transfer coefficients and media of chemical emission) may be of more
35 importance in interpreting modeling results particularly given that the intent of
36 these models are often to be evaluative (screening use) in nature.

37
38 Finally, the Panel supported a more explicit description and justification for the
39 Agency's selection of EFM parameter default values. This information, which should be
40 easily accessible to the EPI SuiteTM user, must provide sufficient detail of the
41 environmental media that the default values purport to represent (e.g., temperate or arid
42 terrestrial system).

1 **iv. Are there other estimation methods that should be considered in**
2 **upgrading EPI Suite™?**

3
4 The Panel was able to identify several innovative methodologies that have the
5 potential to enhance both the accuracy and scope of the EPI Suite™ modules. These
6 methodologies include the: a) least squares adjustment of chemical properties approach
7 (Schenker et al., 2006), b) polyparameter linear free energy relationship approach (Goss
8 et al., 2003. Nguyen et al., 2005), and c) the use of molecular polarizability to predict
9 vapor pressure and K_{OA} (Staikova et al., 2004). In addition, the Panel encourages the
10 Agency to partner with other stakeholders to establish a forum (e.g., technical workshop,
11 interagency workgroup, etc.) to evaluate the various methodologies available for mapping
12 chemical domains in support of future (Q)SAR development and innovations in fate
13 modeling.
14

15 **B. Method accuracy and validation**

16 **i. Is the accuracy of the modules in the EPI Suite™ sufficient for its**
17 **various specified uses?**

18
19 EPI Suite™ is a screening tool that supports Agency risk-based decisions
20 regarding new and existing chemicals. EPI Suite™ outputs are generally found to be
21 within an order of magnitude of measured values, an accuracy standard that has been
22 deemed sufficient by the Agency for defensible decision-making at the screening level.
23 Since many users may not recognize the range of accuracy associated with EPI Suite™
24 output, the Panel encourages the Agency to electronically post a detailed disclaimer that
25 clearly identifies the recommended uses of the current version of the EPI Suite™
26 software.
27

28 Although the accuracy of EPI Suite™ varies depending on endpoint, the Agency
29 staff described EPI Suite's design as intended to provide "best estimates," and in the view
30 of some panel members, the screening level models used for assessing exposure are
31 generally designed to be conservative. The reason for this is that, for a screening level
32 assessment, the Agency generally develops estimates that are conservative (protective).
33 Such conservatism minimizes the probability of users making decision errors based on
34 module output. While minimizing false positive decision errors improves the
35 effectiveness with which the Agency uses its scarce resources, minimizing false negative
36 decision errors also establishes greater confidence that Agency decisions based on EPI
37 Suite™ output will be sufficiently protective of the environment.
38

39 Concerning application of EPI Suite™ output, greater transparency in describing
40 the process by which decision errors are considered in regulatory decision-making would
41 more effectively communicate environmental assessment decisions. By explicitly
42 defining the acceptable level of false negative and false positive decision error rates
43 within each regulatory program that uses EPI Suite™ module output, the Agency would
44 make the basis for its decisions more easily understood.

1
2 In describing EPI Suite™'s level of quality assurance, the Agency confirmed that
3 EPI Suite™ was in full compliance with the EPA's Information Quality Guidelines
4 (USEPA 2002)¹. The Agency has stated that extensive software security precautions
5 have been fully integrated into EPI Suite™ to prevent the possibility of unauthorized
6 algorithm modification. Moreover, the use of scientifically defensible (Q)SARs within
7 the individual modules ensures that the software output is presented in a complete and
8 unbiased manner. The three basic steps employed by the Agency in developing the EPI
9 Suite™ software include the following:

- 10
- 11 • Model Development: This step includes: a) defining the Agency
12 programmatic needs, b) scientific evaluation of the peer-reviewed
13 literature, c) developing and testing the theoretical concept that supports
14 the model and d) developing and documenting the (Q)SAR(s).
 - 15
 - 16 • Model Evaluation: This step includes: a) evaluating the (Q)SAR(s) and
17 their intermediate output, b) evaluating the model results against peer-
18 reviewed measurement data, c) providing basic quality assurance/quality
19 control checks, d) alpha testing the model to ensure that it performs as
20 designed, e) beta testing the model by independent users and f) facilitating
21 peer review of the QSAR by the scientific community.
 - 22
 - 23 • Model Application: This step includes evaluating and documenting the
24 data quality and model performance limitations to ensure that users will
25 apply the model appropriately.
 - 26

27 At the present time, there are relatively few systematic evaluations of the training
28 data sets for EPI Suite™ modules. The Panel strongly recommends that the Agency
29 establish a data quality oversight program that monitors, critically evaluates and
30 incorporates new peer-reviewed measurement data as well as new modeling approaches.
31 Several innovative methodologies offer potential opportunities to improve the accuracy
32 and broaden the scope of EPI Suite™ software. These include the:

- 33
- 34 • least squares adjustment of chemical properties approach (Schenker et al., 2006),
 - 35
 - 36 • polyparameter linear free energy relationship approach (Goss et al., 2003. Ngyuen
37 et al., 2005), and
 - 38

¹ As described in the Council for Regulatory Environmental Models Guidelines (USEPA 2003), EPA's Information Quality Guidelines (USEPA 2002) define quality as a broad-term that includes the concepts of integrity, utility, and objectivity. The Guidelines state that "integrity refers to the protection of information from unauthorized access or revision to ensure that it is not compromised through corruption or falsification. In the context of environmental models, often integrity is most relevant to protection of code from unauthorized or inappropriate manipulation. Utility refers to the usefulness of the information to the intended users. Objectivity involves two distinct elements, presentation and substance. Objectivity includes whether disseminated information is being presented in an accurate, clear, complete and unbiased manner. In addition, objectivity involves a focus on ascertaining accurate, reliable and unbiased information."

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- 1 • use of molecular polarizability as a predictor of physical-chemical properties
2 (Staikova et al., 2004).
3
4

5 EPI Suite's™ data quality should be evaluated at regular intervals (e.g., at least
6 annually). Updates to individual modules should be documented for technical comment
7 and use by the user community. Currently, the Agency has other software packages
8 (e.g., SPARC) at its disposal whose output may be compared to selected output from EPI
9 Suite™.

10 For EPI Suite™ users, the following quality assurance information would be
11 helpful in evaluating and characterizing individual module output:
12

- 13
- 14 • Provide a detailed description of the module chemical training set domain.
 - 15
 - 16 • Flag output when the chemical and associated physical-chemical properties are
17 outside the training set domain.
 - 18
 - 19 • Furnish the range of experimental data used in the module chemical training set in
20 addition to the selected value used in calculations.
 - 21
 - 22 • Provide statistical comparison of results using estimated and experimental data.
 - 23
 - 24 • Identify any chemical fragments that are not captured by the Simplified Molecular
25 Input Line Entry System (SMILES) algorithm within the module output.
 - 26
 - 27 • Identify those chemicals or class of chemicals that have been placed on the
28 'potential problem' list under Toxic Substances Control Act (TSCA).
 - 29
 - 30 • Within the help files, module accuracy or method error should be fully discussed.
 - 31
 - 32 • A description of how default parameters or data were selected should be provided.
 - 33

34 The Panel recognizes the importance of the availability of high quality, peer-
35 reviewed measurement data as the basis for EPI Suite modules. Therefore, the Panel
36 encourages the Agency to upgrade the current set of EPI Suite™ modules to include as
37 much peer-reviewed measurement data of a credible and known quality as possible and
38 remove, where justified, data of lower or unknown quality. Moreover, the Agency
39 should develop a programmatic framework that would facilitate the systematic evaluation
40 of data quality obtained from both intra-Agency and inter-Agency sources. The goal of
41 these activities is to develop improved chemical data training sets of known quality for
42 each of the properties estimated by EPI Suite™. More detailed information can be found
43 in APPENDIX 1Bi.
44
45

ii. Have the modules been adequately validated, and have they been published in the peer-reviewed technical literature or elsewhere?

While no module is ever completely validated, the Panel agreed that the EPI Suite™ modules have, for the most part, been satisfactorily evaluated. The scientific underpinnings of each of the compartment modules have been appropriately vetted in the peer-reviewed scientific literature and the physical-chemical property (Q)SARs have been found to satisfy the OECD principles for (Q)SAR validation. The five OECD principles established for (Q)SAR validation (OECD 2004) are summarized as follows:

- Principle 1: Defined endpoint
- Principle 2: Unambiguous algorithm
- Principle 3: Defined domain of applicability
- Principle 4: Appropriate measures of goodness of fit (e.g., coefficient of determination – R^2)
- Principle 5: Mechanistic interpretation.

OECD Principle 1 requires that (Q)SARs should have a defined endpoint. Most EPI Suite™ modules conform to this requirement. The end point for the biodegradation module (BIOWIN) is less clear because certain aspects of the module (e.g., primary degradation) could range from a minor change in chemical structure (e.g., loss of one halogen, change from one unsaturated to saturated bond in a complex structure) to full mineralization of the chemical. The user should fully recognize that, because of the inherent complexity of the degradation process, ascribing a consistent primary degradation endpoint under all possible environmental conditions may not be feasible. Some panelists commented on the inconsistency in the underlying training data used for calibration of the BCFWIN module (e.g. inclusion of studies involving both parent substance as well as non-parent specific radiotracer studies).

OECD Principle 2 has been consistently achieved by the EPI Suite™ (Q)SARs. Most EPI Suite™ modules are relatively transparent in their design and construction. An overview of their structure and development is provided in the user guide and in the published peer-reviewed literature. The one notable exception to this finding is the biodegradation module (BIOWIN), whose structure and parameterization is less transparent. The Panel strongly recommends that the Agency better define the design, structure and data quality implications of the BIOWIN module. Definition of the environmental medium to which the BIOWIN module output results apply would be a valuable first step. Furthermore, the scientific justification for the scaling rules used to extrapolate results from BIOWIN estimates associated with aqueous environments to soil and sediment should be fully described in the Help files. Finally, the Agency should fully describe the sensitivity of module output when chemical removal through various abiotic processes is prevalent (e.g., sorption, hydrolysis, chemical oxidation, etc.).

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1
2 EPI Suite™ modules are generally consistent with OECD Principle 3. However,
3 the Panel noted that module predictions are less reliable for chemicals that are outside of
4 the chemical training set domain. Moreover, for modules that have multidimensional
5 interpolation domains (i.e., models that use atom/fragment components, e.g., KOWWIN),
6 determining the actual interpolation domain is not trivial.

7
8 A recently peer-reviewed publication evaluated the domain of the chemical
9 training data set utilized by KOWWIN. This work proposes a novel approach for
10 defining the multi-dimensional space that describes the chemical data training set
11 (Nikolova-Jeliazkova, et al. 2005). The Panel encourages the Agency to explore this and
12 other scientific approaches suitable for defining the chemical training set domains for EPI
13 Suite™ modules. The ultimate goal, of course, is to develop a scientifically defensible
14 process by which chemicals are selected for inclusion in the chemical training set
15 domain. Moreover, based on the insight developed through this approach, priorities can
16 be established to target new data collection that efficiently expands the model domain for
17 substances of regulatory importance.

18
19 In general, the EPI Suite™ modules are consistent with OECD Principle 4.
20 External evaluation of an EPI Suite™ module using query chemicals with known
21 properties is the standard procedure for assessing (Q)SAR reliability. External
22 evaluation has produced adjusted R^2 values of approximately 0.75, a value that is
23 considered satisfactory for regulatory screening level chemical evaluation. A few of
24 the EPI Suite™ modules (e.g., BCFWIN, HYDROWIN, etc.) do not appear to have had
25 external evaluation. The Panel strongly encourages the Agency to scan the peer-
26 reviewed literature to determine if external evaluation of these modules has occurred and,
27 if so, is the data quality suitable for supporting upgrades to EPI Suite™.

28
29 Those EPI Suite™ modules which are not regression-based routines do not
30 conform to OECD's Principle 5. However, the EFM modules are mechanistically based
31 and are adequately described in the Help files.

32 **iii. Are some modules more accurate/better validated than others, and if**
33 **so, which need more work?**

34
35 Most of the EPI Suite™ modules have been evaluated sufficiently to support
36 regulatory decision-making. However, all modules would benefit by improved domain
37 mapping, which would allow, amongst other things, the ability of the user to determine *a*
38 *priori* the suitability of a particular module to reliably estimate a given physical-chemical
39 property for a specific chemical.

40
41 Of the EPI Suite™ modules that require additional validation/evaluation beyond
42 that already discussed in the response to the preceding question, the organic carbon
43 partition coefficient model, PCKOCWIN is a priority because this module was developed
44 twenty years ago (1986) and has yet to be revised. Presently, K_{OC} estimation routines use
45 molecular connectivity indices (MCIs) and correction factors based on structural features

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1 of the chemical. MCIs are generally not widely used or accepted by (Q)SAR developers
2 because MCI mechanistic information is difficult to interpret. Finally, the database of
3 K_{OC} values used to develop the present version of the PCKOCWIN module is not as large
4 and inclusive as for other EPI Suite™ modules.

5 **iv. To the extent that modules work together to generate estimates, do**
6 **they do so correctly?**

7
8 EPI Suite™ modules work together to generate scientifically defensible estimates
9 of the physical-chemical properties of chemicals. However, the transfer of data between
10 modules requires further refinement. The Panel encourages the Agency to explicitly
11 describe the protocol (and hierarchy) that govern the passing of physical-chemical
12 property module output to the chemical fate and transport modules. For example, the
13 user may want to know whether a measured physical-chemical property value is used
14 preferentially over a chemical property module prediction in fate and transport modules
15 and the implications of either choice (e.g., advantages of using presumably more accurate
16 measured data over the advantage of using an internally consistent set of physical-
17 chemical properties when estimating chemical fate, (e.g., Beyer et al.2002)).

18
19 To improve transparency in describing module interaction, module inputs as well
20 as outputs should be provided as part of the EPI Suite™ results. Moreover, the Panel
21 strongly supports separating the physical-chemical property estimation modules from the
22 fate modules, such that the fate modules can be executed independently. With respect to
23 module default values for certain parameters (e.g. mass transfer coefficients, media
24 compartment volumes, deposition parameters), the Panel endorses greater user-
25 customization capabilities including the option for batch mode processing with user-
26 defined inputs.

27
28 The Panel found that for some modules, inconsistent results can be obtained for a
29 homologous series of compounds where predictions rely on values for other PER
30 parameters in EPI Suite™. For example, the estimated BCF values for five compounds in
31 the n-alkane series, based on either experimental or predicted log K_{ow} values are given
32 below.

33
34 **Table X: Octanol-Water Partition Coefficients and Estimated**
35 **Bioconcentration Factors for Several n-Alkanes Derived from EPI Suite**
36

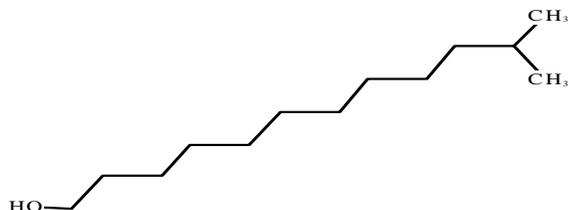
Compound	Log K_{ow} *		BCF
	Experimental	Predicted	
n-octane	5.18	4.27	1944
n-nonane	NA	4.76	93
n-decane	5.01	5.25	144
n-undecane	NA	5.74	528
n-dodecane	6.10	6.23	314

37 *Bolted values used by EPI Suite to predict BCF.
38

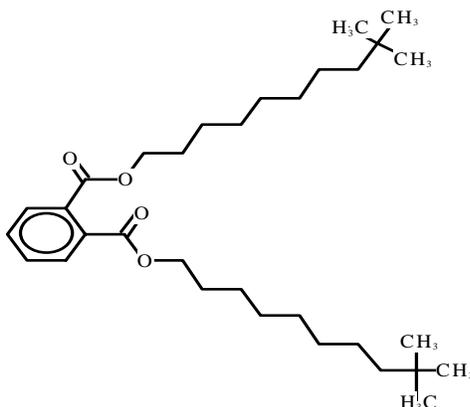
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1 As seen above, the predicted log K_{ow} values show a predictable pattern of
2 increasing hydrophobicity with increasing chain length. However, BCF values do not
3 show this pattern – the shortest chain compound (n-octane) with the lowest predicted log
4 K_{ow} (and an experimental value intermediate between two other experimental values for
5 higher molecular weight alkanes) produces the highest predicted BCF. This pattern is not
6 undone by manually entering an experimental value – for example, entering a log K_{ow} of
7 5.18 for n-nonane gives a predicted BCF (based on that value) of 194, still an order of
8 magnitude lower than the predicted BCF for n-octane, with an identical experimental log
9 K_{ow} . It appears further work may be needed in development and use of correction factors
10 employed to estimate BCF in EPI Suite.

11
12 The common option that allows the user to enter the CAS number of a chemical
13 to obtain the corresponding SMILES string is a convenient feature of all EPI Suite™
14 modules. However, it appears that a number of commercial substances that are not
15 unique structures (i.e., Unknown, Variable Composition and Biologicals - UVCB) are
16 included in the database as single representative structures. There are two principal
17 concerns with this approach. First, it is unclear from the user guide how representative
18 structures have been selected. Second, it is uncertain if predictions derived from unique
19 structures can be reliably extrapolated to characterize the actual complex substance. To
20 illustrate this concern, the representative structure for CAS number 68526-86-3
21 (Alcohols, C11-14-iso-, C13-rich) is shown below.



22
23
24 This isomeric alcohol mixture is reacted with phthalic anhydride to produce CAS
25 number 68515-47-9 (1, 2-Benzenedicarboxylic acid, di-C11-14-branched alkyl esters,
26 C13-rich)



1
2
3 The representative structures selected for these two chemicals are inconsistent
4 since they reflect different alkyl chain branching. Moreover, such arbitrary differences in
5 selection of representative structures can yield misleading predictions for some key
6 endpoints (e.g. biodegradation).

7
8 To address this concern, the user could first be alerted by EPI SuiteTM to the fact
9 that the chemical under consideration is complex and may not have a unique structure
10 and that physical-chemical property predictions may be less certain than for a unique
11 chemical.

14 C. Estimation Methods and Alternates

15 i. Are the estimation methods in the EPI SuiteTM up-to-date and generally 16 accepted by the scientific community for its various uses?

17
18 In general, the Panel concluded that the current estimation methods used in the
19 EPI SuiteTM modules are generally accepted by the scientific community. However, the
20 methods are at risk of becoming outdated as data and practice advance, particularly with
21 regard to the data included in the module training sets. For this reason, the Panel
22 encourages the Agency to evaluate whether the incorporation of newer statistical
23 approaches (e.g., logistical modeling) would increase the accuracy of module prediction.
24 A detailed summary of the relevance and general acceptability of EPI SuiteTM estimation
25 methods is provided in the following bullets.

- 26
27 • *Up-to-date*: The underlying data and statistical models are generally not up to
28 date. The Agency should consider incorporation of new data sets and newer
29 statistical analysis tools to optimize the accuracy of the modules. Linear
30 regression may not always be the optimal statistical model for physical-chemical
31 property estimation.

- 1 • *Acceptance by the scientific community:* Those in the scientific community who
2 understand the role and accuracy limitations of screening models used in
3 regulatory decision-making generally accept the EPI Suite™ module results for
4 many classes of organic chemicals. The EPI Suite™ modules are also generally
5 accepted among regulators. EPI Suite™ modules have been accepted by the
6 OECD and are being tested for implementation in relation to high production
7 volume (HPV) chemicals and the Globally Harmonized System (GHS) for
8 classification and labeling of chemicals by OECD. At the request of the United
9 Nations Sub-Committee of Experts on the GHS, the OECD is developing
10 proposals for classification criteria and labeling of chemicals according to the
11 health and environmental hazards they may present. A Task Force on
12 Harmonization of Classification and Labeling has been established to coordinate
13 the technical work carried out by the experts. OECD typically assigns a
14 reliability code of 2 (valid with restrictions) to EPI Suite™ estimates. Moreover,
15 the extensive peer-reviewed documentation that supports the use of EPI Suite™
16 (Q)SARs as well as the large number of evaluation (validation) studies published
17 demonstrates that EPI Suite™ complies with EPA information quality guidelines
18 (USEPA 2002).

- 19
20 • *Use in assessments:* Within the wider scientific community there is some
21 confusion about whether EPI Suite™ module output is appropriate for full risk
22 assessment or hazard assessment. However, in general, those experts that
23 understand that the EPI Suite™ modules are evaluative by design, hypothesis
24 generators, and first tier predictions of a chemical's fate when the alternative is no
25 data at all support the predictive functionality that the modules provide. More
26 detailed information can be found in the APPENDIX for 1Ci.
27
28

29 **ii. Are there other estimation methods that should be considered in**
30 **upgrading EPI Suite™?**

31
32 Owing to the breadth of this charge question, the Panel's response was two-fold.
33 The first part of the Panel's response is focused on estimation methods that are applicable
34 primarily to new physical-chemical properties (i.e., those that are not currently available
35 within EPI Suite™). The second part of the Panel's response describes the development
36 of methods/approaches that could be used to more effectively estimate properties that are
37 currently available in EPI Suite™.
38

39 With respect to new additional physical-chemical properties, the Panel identified
40 the following as important for expanding the accuracy and scope of EPI Suite™ for
41 organic compounds:
42

- 43 • pKa
44 • Influence of pKa on other physical-chemical properties

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- 1 • Temperature dependency of all physical-chemical properties
- 2 • K_{AW}
- 3 • K_{OA}
- 4 • Bioaccumulation factors for root plants, leaf plants, fish and terrestrial organisms
- 5 (e.g. meat and milk transfer factors)
- 6 • Diffusion coefficients in various environmental media
- 7 • Metabolism and production of stable chemical intermediates
- 8 • Neutral hydrolysis
- 9 • Activity coefficients
- 10 • Sub-cooled liquid vapor pressure and aqueous solubility
- 11 • Surface tension
- 12 • Anaerobic biodegradation potential
- 13 • Ozone depletion potential, greenhouse gas potential, and maximum incremental
- 14 reactivity (MIR) for assessing ozone formation potential.
- 15

16 With respect to EFMs for wastewater treatment, EPI Suite™ currently includes
17 predictions for only a default conventional activated sludge system. Future
18 enhancements should provide options for user-defined treatment systems (e.g., tank
19 dimensions, fine versus coarse bubble diffusers/differs) as well as alternate treatment
20 designs (e.g. aerobic lagoons).

21
22 Several panel members offered the following list of additional chemical properties
23 specifically related to the toxicity and fate of polymers that the Agency may consider
24 adding to EPI Suite™:

- 25
- 26 • Glass transition temperature
- 27 • Crystal melt transition temperature
- 28 • Elastic mechanical properties like bulk modulus
- 29 • Viscosity measures
- 30 • Heat capacity
- 31 • Cohesive energy
- 32 • Flammability
- 33 • Parameters (e.g., degradation rates) influencing environmental persistence
- 34

35 With regard to improving the accuracy of predictions of those physical-chemical
36 properties currently available within EPI Suite™, the Panel identified the following new
37 approaches:

- 38
- 39 • The Agency should consider the use of poly-parameter linear free energy
- 40 relationships (poly-parameter LFERs) and neural networks in module
- 41 optimization as well as partial least squares and support vector machine
- 42 methodologies in data fitting.
- 43
- 44 • In those cases where multiple modules exist that are capable of predicting the
- 45 value of the same physical-chemical property, consensus modeling should be

1 conducted. If all modules for estimating a given property for a particular
2 chemical agree, there is a high level of confidence associated with the property
3 estimation. Conversely, if the modules results vary widely, the reliability of the
4 property prediction is uncertain.

5

- 6 • To the extent that the Agency can document data quality, the Agency should
7 consider moving from two dimensional to three dimensional chemical structure
8 based methods.

9

10 Additional comments relating to this topic can be found in section 1-A-i above.

11

12

13 **2. Functionality**

14 **A. How convenient is the software and does it have all the necessary** 15 **features?**

16

17 Although the software is convenient to use, significant improvements should be
18 made to enhance the appearance, navigability and quality of technical support provided
19 by the EPI SuiteTM software. The following bullets summarize the technical
20 recommendations.

21

- 22 • Currently, the individual property estimation and fate modules cannot be launched
23 from the EPI SuiteTM interface. The Panel supports greater program flexibility
24 that would allow software users the ability of launching individual modules
25 directly from the user interface, with appropriate indication of options for entering
26 data or utilizing EPI SuiteTM-generated values necessary to run the modules.

27

- 28 • To ensure that software users are cognizant of the quality assurance limitations
29 associated with module output, individual modules should alert the user when a
30 chemical's physical-chemical properties are outside the chemical training set
31 domain.

32

- 33 • Although EPI SuiteTM operates on a WindowsTM platform, the graphical user
34 interface (GUI) has an archaic DOS appearance. The Panel encourages the
35 Agency to upgrade EPI SuiteTM's GUI to reflect a more typical WindowsTM
36 operating system environment.

37

- 38 • To minimize the loss of data when new versions of EPI SuiteTM are released, the
39 Panel recommends that the new version installation program not delete chemical
40 data input by the user but, rather, only overwrite older versions of EPI SuiteTM
41 software itself.

42

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- 1 • To address the myriad of data reporting requirements, the Panel recommends that
2 users have the option of saving output files in various formats (e.g., WordTM,
3 WordPerfectTM, ExcelTM, etc.).
4
- 5 • Providing greater flexibility for inputting data files in batch mode e.g., provision
6 of a screen that allows EPI SuiteTM users the ability to simply “cut and paste”
7 Chemical Abstract Services Registry Number (CAS) numbers or SMILES
8 notations would increase efficiency.
9
- 10 • EPI SuiteTM EFM module users would benefit from having access to a simple
11 flow chart that clearly describes the data processing steps that result in generating
12 environmental fate model output.
13
- 14 • To enable users to access various data sets simultaneously, the EPI SuiteTM
15 program should allow minimization of all screens.
16
- 17 • To reduce confusion when saving a chemical name run (via Save User), it would
18 be helpful if the program used as a default the full chemical name (or a truncated
19 version), rather than the most recently saved name.
20
- 21 • To improve program navigability, all parameters should be located in a single
22 location rather than having some parameters placed in the "Functions - Other"
23 category.
24
- 25 • The default option for displaying module results should be the full output results
26 category rather than simply furnishing the summary output results.
27
- 28 • Use of color-coded text to distinguish experimental values from predicted values
29 or to alert users of chemicals whose properties were outside those contained in the
30 module’s chemical data training set would help to minimize misinterpretation of
31 results.
32
- 33 • When inputting a chemical based on SMILES notation alone, the chemical name
34 should be displayed in both the data entry screen and in the output file.
35
- 36 • In the AOPWIN module, EPI SuiteTM should specify the environmental
37 conditions that are associated with the default concentrations of hydroxyl radical
38 and ozone and allow user input of alternative hydroxyl radical and ozone
39 concentrations.
40
- 41 • Clarify the units used in the EPI SuiteTM module PCKOCWIN.
42
- 43 • BOWIN Help information should clearly state the conditions which pertain to
44 this program’s estimates (e.g. aqueous slurry) as well as decision rules for
45 extension of BOWIN results to other media (e.g., sediment, soil).
46

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- 1 • More details regarding the structure, function and parameterization of the
2 WVOLWIN module should be provided in the Help files. For example, it is
3 unclear what default values are being used for air and water temperature, water
4 advective flow, depth of water etc.
5
- 6 • For the sewage treatment plant module, i.e., STPWIN, the Help files fail to
7 provide the default plant operating conditions. Temperature of water, whether
8 the plant has only secondary treatment or includes tertiary treatment as well, solid
9 retention time for the activated sludge systems etc. should be provided in the Help
10 files.
11
- 12 • Since AOPWIN and the Level 3 fugacity module output is sensitive to mass
13 transfer rates as well as degradation/transformation rates, the default values (and
14 their associated temperature dependency) should be provided in the Help files or
15 in an appendix in the user guide.
16
- 17 • Experimental data that may be available for a specific structure is not provided for
18 some endpoints (e.g. BIOWIN, BCFWIN).
19
- 20 • Entering air advection times in hours is not intuitive. Users should have the
21 option of entering wind speed instead.
22
- 23 • On the K_{oc} tab, it is impossible to determine whether the module uses the K_{ow}
24 method, as K_{ow} is not a calculated property in the results.
25
- 26 • In EPI SuiteTM module results, it would be preferable to list experimental values
27 in the same order as predicted values are given (i.e., boiling point, melting point,
28 and vapor pressure).
29
- 30 • For the example of lindane, there seems to be a problem with experimental results
31 for melting point and boiling point (i.e., values in wrong order).
32
- 33 • In the half-life selection module (LEVEL3NT) the user is not allowed to specify a
34 model estimate or a selected value for air, which is an option for the other
35 environmental media.
36
- 37 • For those physical-chemical properties for which two or more methods are
38 currently available, EPI SuiteTM should provide to the user the ability to select
39 which module they would prefer to use (e.g., water solubility).
40
- 41 • The reference feature should be enhanced by allowing the user to easily access
42 individual references (including a brief abstract) through addition of a simple pop-
43 up window.
44
- 45 • For key references, EPI SuiteTM should provide links to web pages where pdf
46 versions of the documents can be accessed, if available.

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- 1 • The Help files should contain the list of all references used in developing the
2 predictive models.
- 3
- 4 • When modeled property estimates are passed on to other modules (e.g., fate and
5 transport modules), the EPI Suite™ program should identify to the user the values
6 that are passed as well as provide clear documentation in the user guide of the
7 protocol used to establish data transmission priority. This is especially important
8 when there is more than one method available for estimating a particular property,
9 e.g., Henry's law constant.
- 10
- 11 • Where the EPI Suite™ Help files explicitly indicate that certain chemicals have
12 been excluded in the database (e.g., CAS Number database), supporting
13 explanation should be provided.
- 14
- 15 • Help files and other documentation should be regularly checked by the Agency
16 for typographical errors.
- 17
- 18 • The Agency should consider adding a "comments" facility to the EPI Suite™ to
19 enable receipt and incorporation of feedback from users such as identification of
20 errors and recommendations.

21

22 **B. Are there places where EPI Suite™'s user guide (and other program**
23 **documentation) does not clearly explain EPI's design and use? How can**
24 **these be improved?**

25

26 The user guide should more clearly identify the modules which can be executed
27 independently and the features available for a particular module when executed alone.
28 The stand alone modules could be identified in a separate highlighted section. Some
29 features are unavailable when executed as part of EPI Suite™ yet can be accessed in
30 stand alone operations. The "Experimental Value Adjusted" option in KOWWIN and
31 HENRYWIN is an example. In addition, separate sections in the user guide should
32 incorporate increased discussion of training set domains and uncertainty in predictions, as
33 noted elsewhere in this report.

34

35 The Panel agreed that the EPI Suite™ user guide provided a clear and succinct
36 description of the design and use of the software. However, the Panel noted that the
37 documentation quality was uneven with many sections supported by detailed references
38 while others were noticeably devoid of such support. Moreover, the Panel was
39 unanimous in its recommendation that the EPI Suite™ software should allow users the
40 ability to easily download and print a copy of the user manual as a stand-alone document.

41

42 With respect to general improvements for the user guide, the Panel recommends
43 that the Agency develop a separate detailed guide for activities or functions common
44 among the various modules (e.g., how to import chemicals through the SMILES notation,

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1 function keys and buttons, use of results and structure windows, etc.) as well as a quick
2 start guide for experienced users. Finally, the guide should clearly describe those
3 modules that predict chemical properties based on the output from other modules (e.g.,
4 use of K_{ow} output to predict bioconcentration factors through BCFWIN.

5

6 **C. Are there aspects of the user interface (i.e., the initial, structure/data**
7 **entry screen; and the results screens) that need to be corrected, redesigned,**
8 **or otherwise improved? Do the results screens display all the desired**
9 **information?**

10

11 The Panel applauds the multi-faceted functionality of the EPI Suite™ user
12 interface. However, the Panel is of the unanimous opinion that EPI Suite™ does not take
13 full advantage of the opportunities provided by a Windows™ environment. Moreover,
14 while there are many positive features associated with the EPI Suite™ user interface
15 including its documentation and HELP file availability, there are also opportunities for
16 substantial improvement. Recommendations for improving the overall functionality of
17 the user interface could include the following:

18

- 19 • The format for module output should be user defined and include the following
20 Windows™-based display options: Excel™, WordPerfect™ and/or Word™ file.
- 21
- 22 • When multiple measured values are available within a module, the user should
23 have the option to select which measured value is applied in the calculations.
- 24
- 25 • Under the fugacity tab, the input screen should identify the source of module
26 input(s) as well as what algorithms are being executed.
- 27
- 28 • Because a user can enter data through either a SMILES string or the chemical
29 name, the screen could more clearly indicate that both options are possible.
- 30
- 31 • The “previous” button has limited functionality and does not seem to work in all
32 scenarios. The “previous” option should allow the user to return to a chemical
33 when evaluating multiple chemicals and ideally recall more than simply the most
34 recent chemical evaluated.

35

1 **D. Currently one enters EPI SuiteTM using SMILES and CAS; are there**
2 **other ways to describe the structure (e.g., ability to input a structure by**
3 **drawing it), that should be added?**

4
5 The SMILES structure and Chemical Abstract Services (CAS) registry number
6 input options are adequate to describe and query chemical structures. However, the
7 addition of an input drawing program would extend the utility of the EPI SuiteTM to users
8 who are unfamiliar with the SMILES notation. Alternatively, the user could be directed
9 towards commercial packages to assist in the derivation of SMILES structures.

10
11 When CAS registry numbers are unavailable, users typically prefer to draw their
12 structures rather than use a string language input. Moreover, the use of SMILES may
13 limit the users of EPI SuiteTM to those with basic knowledge of organic chemistry.

14
15 Inclusion of a two-dimensional [2D] structure drawing program in addition to
16 SMILES will be valuable to users with limited knowledge of organic chemistry. It is
17 also useful to highlight to current users that a structure drawn in commercial software
18 packages (e.g., Cambridge Soft 's ChemdrawTM) can be copied and pasted directly into
19 EPI SuiteTM.

20
21 The Panel does not recommend that the Agency attempt to develop its own
22 structure drawing program, but, rather purchase/license one of the many commercially
23 available software packages. There are several computer-based chemical drawing
24 packages that generate SMILES or other 2D [and 3D] structure tables. The Panel noted
25 the following observations that support utilizing commercially available software
26 packages:

- 27
28 • Most commercially available software packages are generally accepted by the
29 scientific user community.
30
31 • Programs like those offered by Elsevier's MDL and ChemdrawTM have options to
32 execute batch mode operations as well as read and write structure files interfaces
33 to other commercial software.
34
35 • The MDL software package has a module that effectively models chemical
36 properties of linear polymers.
37
38 • Both the MDL (1) and ChemdrawTM (2) software packages can effectively draw
39 isomeric structures and have the ability to interface to three dimensional [3-D]
40 chemical structure generating programs.

1 **E. EPI Suite™ has many convenience features, such as the ability to**
2 **accept batch mode entry of chemical structures, and automatic display of**
3 **measured values for some (but not all) properties. Are there other features**
4 **that could enhance convenience and overall utility for users?**

5
6 While there are a number of features in EPI Suite™ that increase the convenience
7 of the program (including multiple modes of identifying a chemical of interest in the
8 input, and allowing for user-specified input parameters), the Panel has recognized that the
9 program interface should balance convenience with transparency including
10 characterization of the uncertainty associated with model output. In other words, while
11 the Panel is cognizant of the importance of usability in executing the EPI Suite™
12 programs, convenience should not come at the expense of providing users with a better
13 sense of how estimated values are derived.

14
15 The following bullets summarize the specific Panel recommendations with
16 respect to additional features to enhance software convenience.

- 17
18 • To encourage examination of the sensitivity of user input on module output, the
19 Panel recommends that the user have the ability to execute the fate modules
20 separately from the physical-chemical parameter prediction models (with the
21 caveat noted previously in Section 2A).
22
23 • The CAS number database should be validated in the current version (in
24 particular, discrepancies between CAS numbers and SMILES notation), and
25 regularly updated with new information.

26
27 There is a discrepancy between the number of chemicals for which SMILES
28 notation exists and the number of chemicals in the TSCA inventory. It would be
29 valuable to document within the SMILES HELP files the reason for the difference
30 in chemical coverage. The documentation on SMILES refers to approximately
31 20,000 discrete organic chemicals in the original TSCA inventory that are in the
32 SRC database, while the June 2005 U.S. Government Accountability Office
33 (GAO) report references 62,000 organic chemicals in the original inventory
34 (GAO, 2005).

- 35
36 • For the batch mode entry feature, the system should allow user-specified inputs of
37 physical-chemical properties.
38
39 • Rather than having the module output written to the directory containing the
40 program for the batch mode entry feature, it would be preferable to give the user
41 the option of naming the output file and identifying the location where it will be
42 saved.
43
44 • The output data in batch mode should include CAS numbers for each chemical as
45 well as the names and SMILES notation.
46

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- 1 • The Name Lookup feature should be added to each individual module.
- 2
- 3 • For chemicals that have isomers, the module output should explicitly state that
- 4 fact. The output display should include identification of the other isomers that
- 5 exist, by name and CAS number.
- 6
- 7 • For results displayed in summary format, measured values should be given for
- 8 several isomers of the chemical assessed, if available.
- 9
- 10 • For both the summary and full output options, repeating the listing of
- 11 experimental values in the results screen for a given parameter is confusing and
- 12 should be avoided (e.g., experimental aqueous solubility in both fragment
- 13 approach and log K_{ow} approach).

14

15 **F. Are property estimates expressed in correct/appropriate units?**

16 In general, the Panel found no specific concerns regarding the units used to

17 express the property estimates. However, the Panel has made the following

18 recommendations that should improve the overall utility of module output.

19

- 20
- 21 • Output data should be presented in International System of Units (SI) units.
- 22
- 23 • There should be consistency with the use of significant figures.
- 24
- 25 • For BCFWIN the units are L/kg (wet weight) for fish and should be included in
- 26 the output.
- 27
- 28 • Units should be specified for log K_{oc} .
- 29
- 30 • The Agency should provide a unit conversion program to allow the user the
- 31 option to convert from one set of output units to another.

32

33 **G. Is adequate information on accuracy/validation conveyed to the user**

34 **by the program documentation and/or the program itself?**

35 In general, the Panel found that the information on module accuracy and

36 validation was conveyed adequately to the user, but not in a consistent and transparent

37 manner. For the sake of clarity, the Panel has addressed accuracy/validation issues

38 pertaining to (Q)SARs (i.e., algorithms) and the actual property estimation outputs

39 separately.

40

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i) Is adequate information on accuracy/validation conveyed with respect to the QSAR predictive model itself?

The Panel found that, while regression statistics are provided in assessing model performance and residual error, for most modules, this information is not generally transparent to the user. Moreover, when such information is available, it is uncertain as to which version of the module the reported analysis applies. It should be straightforward to determine a common set of statistical significance measures valid across all modules that would provide common and comparative measures of accuracy and validity. A suggested set of module performance metrics for consideration include:

V.x.y =	model version
N(T) =	number of compounds in the training set
N(O) =	number of outliers removed in developing the module
R ² =	standard coefficient of determination
Q ² =	leave-one-out cross-validation coefficient.
SD =	standard deviation of fit
R(l) =	lower value of the range of the property in the training set
R(u) =	upper value of the range of the property in the training set
MRE =	mean residual error
SRE =	standard deviation of residual error
R _X =	average correlation coefficient for models built from X random values of the dependent variables contained in the training set
R(t) _X ² =	correlation coefficient of an external validation set of X compounds

In addition, for each of the endpoints predicted by QSAR, a brief discussion on the measured error associated with current test protocols would be valuable. Any insights regarding trends in measurement error (e.g. measurement error of Log K_{ow} and water solubility tends to increase with increasing Log K_{ow} or decreasing water solubility, respectively) should be summarized.

ii) Is adequate information on accuracy/validation conveyed with respect to making the property estimation of a particular test chemical?

The Panel concluded that a major shortcoming of EPI SuiteTM is that the user is given no indication as to whether the domain of module applicability is appropriate for the test chemical. Currently, the decision to use a module for a specific chemical appears to be based on past experience and/or professional judgment. This approach is not transparent and could lead to inconsistency and error in assessments among chemicals. How the Agency uses (or does not use) or interprets EPI SuiteTM results in making decisions is an important consideration in determining if the software provides the degree of accuracy that supports its intended use.

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1 Upgrades to EPI Suite™ provide the Agency with an opportunity to better
2 understand the accuracy of the software's estimates. For example, the Agency could
3 compare earlier estimates to any new measured values that have since been published for
4 the chemical of interest. After an EPI Suite™ PER or EFM has been upgraded, the new
5 estimate can be compared with the earlier estimate for the same chemical. Assuming the
6 more recent upgrade is producing more accurate estimates in general, results from the
7 new predictions can indicate the degree of over- or under-prediction for the parameter of
8 concern in the original assessment. Comparing either new experimental or estimated data
9 to decision-making criteria can be used to assess the performance of EPI Suite™ in
10 supporting regulatory decision-making

11
12 The Panel also endorses an independent "model domain" analysis to improve
13 accuracy and reliability estimates of chemical property values. In this analysis, the
14 degree of molecular similarity of the test chemical to that of the chemicals used in the
15 module training set establishes the reliability of a property estimate. The Panel did not
16 have the necessary expertise to provide specific advice on preferred domain analysis
17 methods but encourages the Agency to seek experts for technical guidance so that this
18 functionality can be included in future EPI Suite™ upgrades.

19 20 **3. Appropriate Use**

21 **A. Currently Identified Uses**

22 **i. Is the science incorporated into EPI Suite™ adequate for each of** 23 **these current uses?**

24
25 All of the modules in EPI Suite™ are generally accepted for use in risk-based
26 priority setting, screening level risk assessment and prioritization for chemical testing, for
27 the chemical classes to which the modules apply. Given the large number of chemicals
28 that the Agency must screen in a short period of time, reliance on (Q)SAR module output
29 is justified. The modules are expected to provide order of magnitude estimates, an
30 accuracy standard that is generally acceptable by the Agency for screening level
31 assessments. This level of accuracy should be clearly conveyed to users outside the
32 Agency.

33
34 The Agency should continue to validate, update, and investigate the uncertainty
35 associated with the modules in various regulatory programs. A more extensive analysis
36 and explanation of the limitations of the PERs and EFMs would help clarify appropriate
37 use.

38 **ii. If not, what improvements are needed to make EPI Suite™ adequate** 39 **and what alternative approach could be used in the interim?**

40
41 There are specific uses of EPI Suite™ that are not entirely appropriate for
42 supporting the PMN and pollution prevention (P2) programs. At present, the chemical

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1 domains that are used by (Q)SARs do not provide adequate coverage of nanoparticles,
2 inorganic compounds, organo-metallic and some polymeric chemicals (as well as other
3 classes of chemicals). Application of (Q)SARs to chemicals outside the domain of the
4 training set is likely to result in unreliable estimates. The Panel recommends that the
5 Agency collect more peer-reviewed measurement data on the physical and chemical
6 properties for these chemicals with the intent of either expanding the domain of the
7 existing (Q)SARs or for creating new (Q)SARs specifically for these classes of
8 chemicals.

9
10 **B. Potential Additional Uses**

11 Given the Agency's global leadership in the field of chemical screening to
12 emerging industrial economies in Asia, South America, Eastern Europe, and Africa, it
13 should come as no surprise that these regions are adopting EPI Suite™ in their regulatory
14 programs as well.

15
16 EPI Suite™, if translated into major foreign languages (e.g. Arabic, Spanish,
17 Portuguese, French, Russian, Standard Chinese and Mandarin, Bahasa Indonesia, and
18 Hindi), is a practical and scientifically-credible risk management technology transfer that
19 will allow countries with emerging industries to establish sustainable chemicals
20 management systems. The United Nations (UN) Strategic Approach to International
21 Chemicals Management (SAICM) project represents an ideal forum in which the benefits
22 of EPI Suite™ application can be shared with the international regulatory community.

23
24 In addition to the direct uses of EPI Suite™ by the Agency, the following
25 additional potential uses have been identified.

26
27 • EPA and other Federal Agencies:

28 EPI Suite™ is clearly seen as an important tool in any regulatory program
29 that evaluates chemicals for public health and environmental safety.
30 Agency programs that benefit from EPI Suite™ include: a) EPA Office of
31 Pesticide Programs (OPP), b) EPA Office of Water (OW), c) EPA Office
32 of Solid Waste and Emergency Response (OSWER) and the c) US Food
33 and Drug Administration.

34
35 • Private Industry:

36 Industrial applications where EPI Suite™ software can be valuable
37 include the development of more environmentally friendly products or
38 "green" engineering processes.

39
40 EPI Suite™ can be used to support the issuance of chemical exposure-
41 based waivers that reduce the use of animal testing under programs such
42 as TSCA and HPV Challenges world-wide.
43

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1 EPI Suite™ output can inform and guide environmental exposure
2 monitoring programs.

3
4 • International Regulatory and other Programs:

5 EPI Suite™ output can be used to support hazard classification when
6 experimental data are not available.

7
8 EPI Suite™ output can be used as part of the process to conduct Persistent
9 Bioaccumulative and Toxic (PBT) identification/categorization.

10
11 EPI Suite™ output can be used to support chemical assessment and
12 management programs especially for High Production Volume (HPV)
13 chemicals.

14
15 EPI Suite™ output can be used to support global initiatives such as the
16 Stockholm Convention to control the long-range transport of Persistent
17 Organic Pollutants (POPs) or other assessments of the potential for long
18 range transport of chemicals and other greenhouseGreen House gas
19 assessments.

20
21 EPI Suite™ may play a significant role in the OECD (Q)SAR ToolBox.

1
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APPENDIX 1: PROCEDURE

1
2
3 In 2006 the EPA Science Advisory Board convened the EPI Suite™ Review
4 Panel to review and evaluate the supporting science, functionality, current uses and
5 potential uses of the EPI Suite™ model. The Panel was formed in accordance with the
6 principles set out in the 2002 commentary of the Science Advisory Board, *Panel*
7 *Formation Process: Immediate Steps to Improve Policies and Procedures* (EPA-SAB-
8 EC-COM-02-003).

9
10 The Office of Pollution Prevention and Toxics (OPPT) in the Office of
11 Prevention, Pesticides and Toxic Substances (OPPTS) provided access to the model and
12 supporting materials, which are routinely available at the EPA website. The Panel held
13 public conference call meetings on February 22 and March 1 to prepare for a face-to-face
14 meeting March 7-9, 2006, in Washington, DC. The conference calls provided an
15 opportunity for the Agency to clarify questions of the Panel. Subsequently, panelists
16 prepared written responses to the charge questions, which were distributed at the face-to-
17 face meeting.

18
19 The primary purpose of the face-to-face meeting was for the Panel to reach
20 consensus on the content of their response to the charge questions, to capture that
21 consensus in writing, to brief OPPT staff on the major findings and conclusions, and to
22 respond to questions. A first draft consensus report was distributed and revised at the
23 meeting. The second draft was distributed to the Panel for further editing and a third
24 draft posted at the SAB's website March 24, 2006 for consideration on a public
25 conference call April 5. The Panel completed its editing by email.

26
27 Following the Panel's approval draft report, it was forwarded to the chartered
28 SAB for review and approval. In its review of the Panel's report, the chartered SAB
29 considered whether the report addresses the charge in a clear and logical manner and
30 whether the Panel's conclusions and recommendations are supported by the body of the
31 report. The Board approved the Panel's report at a public meeting [DATE?]

32

1 **APPENDIX 2: Summary Assessment of EPI Suite™ Core Models**

2 (Note: ECOSAR is not reviewed)

3

Model	Assessment
AOPWIN	Atmospheric oxidation/ozone reaction rates are predicted using AOPWIN using the Atkinson fragment and functional approach method. It is the generally accepted approach for estimating these properties. It has been validated on a relatively small dataset of 77-79 chemicals. EPA should consider more validations for this method. $R^2 = 0.93$
BCFWIN	BCFWIN is generally accepted as the best fit to existing bioconcentration data. BCFWIN does not appear to have been externally validated or the information is not available in the user guides. If these models have been externally validated in the literature by various investigators, EPA should include this data in the user's manuals. No R^2 .
BIOWIN	The (Q)SPR estimation of biodegradation has inherent problems, one of which is the lack of reproducibility of measured biodegradation data. The BIOWIN model is reasonably well accepted and generally performs as well as or better than the available models. EPA should summarize all available validation data for BIOWIN in the users manual so that this information is readily available. Also, EPA should consider giving more advice on which of the 3 BIOWIN model approaches is most appropriate in a given situation. $R^2 = 0.5-0.97$
HYDROWIN	Hydrolysis rates for a specific set of functional groups are predicted by HYDROWIN and are a generally accepted approach. HYDROWIN does not appear to have been externally validated or the information is not available in the user guides. If these models have been externally validated in the literature by various investigators, EPA should include this data in the user's manuals. No R^2 .
KOWWIN	The KOWWIN model is well accepted, uses an accepted fragment-based technique and is an important (Q)SPR for regulatory use. It generally performs better than most existing (Q)SPR Kow prediction methods. The external validation data for this method is good and the summary information is available to the user. $R^2 = 0.94$.
MPBPVP	The MPBPVP (Q)SPR is accepted as a good estimator of BP, MP and VP. The melting point (Q)SPR is the weakest of this group because the external validation coefficient of determination was reported as 0.66. The standard deviation of 63 K is also indicative of some prediction error. It is not likely that a significantly more accurate melting point determination is necessary for EPA regulatory programs and this method should be satisfactory for most regulatory uses. $R^2 = 0.92-0.95$.
HENRYWIN	Uses two different methods and produces two different estimates (bond and group contribution) for air-to-water partition coefficient. The models are generally accepted with $R^2 = 0.94-0.96$.
PCKOCWIN	The as a good estimation tool of soil sorption coefficients (Koc) based on first order molecular connectivity index (MCI). It is satisfactory for most regulatory uses. $R^2 = 0.86-0.96$.
WATERNT	WATERNT uses the atom fragment contribution (AFC) method to predict water solubility building upon the KOWWIN methods water solubility of organic compounds at 25°C is predicted. $R^2 = 0.87-0.98$.
WSKOWIN	WSKOWIN is a good model for prediction of water solubility. It has been validated with a large dataset with a high coefficient of determination, $R^2 = 0.9$.
WVOLVIN	Estimates volatilization half-lives from a model river and lake. The program's default parameters for a model river will yield a half-life that is indicative of the fastest volatilization that may be expected in environmental waters (a shallow, rapidly moving river with strong surface wind). The default parameters for the lake yield a much slower rate. The EPI interface program executes the WVOLNT (Volatilization Rate from Water) program by transferring the Molecular Weight, the Henry's Law Constant, and various volatilization parameters to WVOLNT. No R^2 .

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LEVEL3NT	<p>Half-lives are required for air, soil, sediment and water . . . the fugacity can not run without them. If the half-lives in air, water, soil and sediment are known, the "Use Half-Lives Entered Below" should be selected and the known values should be entered in the appropriate fields. Often, however, these data are not available and require estimation. The BIOWIN and AOPWIN programs are used to make these estimates. The AOPWIN air estimate is based upon estimated hydroxyl radical and ozone rate constants. AOPWIN does have an experimental database containing more than 700 compounds. If an entered structure has a database match, the database value is used instead of the program estimate. The half-life for degradation of a chemical in water, soil, and sediment is determined using the ultimate biodegradation expert survey model of the BIOWIN estimation program. This estimation program provides an indication of a chemical's environmental biodegradation rate in relative terms such as hours, hours to days, days, days to weeks, and so on; the terms represent the approximate amount of time needed for degradation to be "complete". This output cannot be used directly by the level III multimedia mass balance model. The mean value within the estimated time range returned by Biowin3 is converted to a half-life using a set of conversion factors. These conversion factors consider that 6 half-lives constitute "complete" degradation of a chemical substance, assuming first-order kinetics. The resulting conversion factors for water are provided below. The Fugacity Model can not run without a vapor pressure. If the vapor pressure is not user-entered, the model uses the vapor pressure estimate by the MPBPWIN Program. If the MPBPWIN Program estimates a vapor pressure of zero (which can occur if an estimate is less than 1.00e-40 mm Hg), the fugacity model uses an assumed value of 1.00e-15 mm Hg (this value is low enough to have no sensitivity effect in the fugacity estimates). The model also requires a log Kow value. If the log Kow is not user-entered, the model uses the value from the KOWWIN Program (an experimental database value is used if available instead of the estimate). The Fugacity model in EPIWIN has limited user-access to many parameters in the Mackay Level III Model. For example, parameters such as rain rate, aerosol deposition, soil water runoff, and diffusion mass transfer coefficients can not be changed by the EPIWIN user. For these parameters, EPIWIN relies solely upon the defaults values as determined by Mackay and co-workers. This greatly simplifies application of a Level III model for most users. No R².</p>
STPWIN	<p>The STPWIN program is a version of the Toronto Model originally developed by Donald Mackay and colleagues at the University of Toronto. Includes outputs on: Bio P: the biodegradation half-life (in units of hours) in the primary clarifier of a sewage treatment plant (STP). Bio A: the biodegradation half-life (in units of hours) in the aeration vessel of an STP. Bio S: the biodegradation half-life (in units of hours) in the final settling tank of an STP. All STP parameters are now accessed from the main menu bar by selecting "STP". The STP program uses only default operating conditions of a model sewage treatment plant operating at 25 degree C. No R².</p>

1

APPENDIX for 1Ai

1
2
3 The upgrades to EPI Suite™ could include a module containing algorithms for
4 estimating the mass-transfer coefficients (MTCs) used in the EFM Category as well as
5 allowing for user-entered values. A recent study comparing the outputs of five
6 multimedia fate models demonstrated that model homogenization was possible only
7 when the numerical values of the dozen or so MTCs were numerically equal (Cowan, et
8 al., 1995). Otherwise, the computed concentration levels, mass fractions in the
9 compartments and the chemical residence time estimates were dramatically different,
10 many by orders- of-magnitude. Typically the numerical values of these MTCs vary by a
11 factor of ten at a particular environmental interface and sometimes much more
12 (Thibodeaux, 1996).

13
14 The LEV3EPI module for example, contains twelve default MTC values; these
15 were likely chosen by the model developers and are embedded within the code. In
16 addition to having chemical species and physical property dependence, the MTCs are
17 also functions of parameters that characterize the sizes, fluid dynamics, etc., of the
18 environmental compartments. In the future, as EFMs develop in sophistication the users
19 will need the option of having algorithms for estimating MTCs, including those that are
20 most representative of the environmental compartments into which the chemicals are
21 entering.

22
23 It is possible and appropriate for EPA, with only a modest expenditure of
24 resources, to develop estimating algorithms for these MTCs. A sizable quantity of data
25 and accompanying theoretical models exist in diverse types of published literature. In
26 general the tasks required in the algorithm development efforts will include the collection
27 and evaluation of the existing data followed by producing the appropriate theory-directed
28 statistical correlations needed for their estimation. These final algorithms should be
29 similar to those in the PER Category of the EPI Suite™. Some limited compilations of
30 these MTC algorithms are available in textbooks and other documents (Thibodeaux,
31 1996; DiToro, 2005; Trapp and Matthies, 1998). Many are imbedded within existing
32 Agency software, EXAMS for example. However, there is no single location for
33 accessing such parameters for direct use by the Agency or others. By having such an EPI
34 Suite™ module (e.g., MTCWIN) a major input parameter for the LEV3EPI could be
35 definitively selected by the user thereby eliminating one level of uncertainty that
36 presently exists by relying on unknown imbedded default values.

APPENDIX for 1Bi

The following descriptions are edited versions of the accuracy statements in the EPI Suite™ HELP Files.

Estimation Accuracy of **WATERNT**: The statistical accuracy of the current 1000 compound training set is excellent; the correlation coefficient (R^2) is 0.975, the standard deviation is 0.336 and the absolute mean error is 0.28. However, to be effective, an estimation method must be capable of making accurate predictions for chemicals not included in the training set. Currently, WATERNT has been tested on a validation dataset of 3,923 compounds. The validation set includes a diverse selection of chemical structures that rigorously test the predictive accuracy of any model. It contains many chemicals that are similar in structure to chemicals in the training set, but also many chemicals that are different from and structurally more complex than chemicals in the training set. Statistical performance for estimated vs. experimental log WatSol (moles/L) are: $n = 3923$; $R^2 = 0.86$; $sd = 0.869$; $me = 0.70$.

Accuracy of **AOPWIN**: The accuracy of the estimation methods used by the Atmospheric Oxidation Program can be examined by comparing a list of more than 640 experimentally determined hydroxyl radical rate constants to the program's estimated rate constants. Over 90 percent of the estimated rate constants for the 647 different chemicals are within a factor of two of the experiment value. Over 95 percent of the estimates are within a factor of three of experimental. This can be compared to the PCFAP program (Fate of Atmospheric Pollutants) of the USEPA GEMS software which estimates the same rate constants as AOPWIN. For 617 compounds (PCFAP can not estimate or produces program errors for the remaining experimental values), PCFAP is within a factor of two for about 49 percent of the experimental values and within a factor of 3 for about 65 percent. PCFAP is particularly inaccurate for many compounds containing nitrogen, sulfur or phosphorus. The document "Estimation Accuracy of the Atmospheric Oxidation Program" contains a compilation of the experimental rate constants used to determine the accuracy of AOPWIN and PCFAP. Each chemical in the compilation includes the experimental rate constant, the AOPWIN estimate, the PCFAP estimate, and the SMILES notation for that chemical. For Aromatic Compounds, one of the advantages of the SMILES interpreter used by AOPWIN is the ability to identify individual aromatic rings and ring structures. This allows the overall rate constant estimation of many aromatic compounds to begin with an experimentally measured value for the basic ring structure. For example, if 1-methylnaphthalene is entered into AOPWIN, AOPWIN finds the naphthalene ring and assigns it the experimentally measured value for naphthalene ($21.6 \times 10^{-12} \text{ cm}^3/\text{molecule-sec}$). It then adjusts the experimental naphthalene value for one methyl group attachment to an aromatic ring to yield an overall estimate of $56.9 \times 10^{-12} \text{ cm}^3/\text{molecule-sec}$ (the experimental value for 1-methylnaphthalene is 53.0×10^{-12}). AOPWIN identifies and uses the aromatic

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1 rings (15) that have experimental values ($\times 10^{-12}\text{cm}^3/\text{molecule-sec}$) and 7 rings are
2 assigned a value based primarily upon experimentally measured ionization
3 potentials ($\times 10^{-12}\text{cm}^3/\text{molecule-sec}$):
4

5 Accuracy of **BIOWIN**: BIOWIN produces two separate MITI probability
6 estimates for each chemical. The first estimate is based upon the fragments
7 derived through linear regression. The second estimate is based upon the
8 fragments derived through non-linear regression. Prediction accuracy of the
9 training and validation sets are listed below. The validation set is completely
10 independent of the training set. Chemicals in the validation set were not used to
11 derive any fragment values. The numbers correspond to correct predictions (either
12 "readily degradable" or "not readily degradable"):

13 Training Set: Critically Evaluated as "Readily Degradable"

14 Italian (Italy) Linear Model: 201/254 (79.1%)

15 Non-Linear Model: 204/254 (80.3%)

16 Training Set: Critically Evaluated as "Not Readily Degradable"

17 Italian (Italy) Linear Model: 284/335 (84.8%)

18 Non-Linear Model: 284/335 (84.8%)

19 Training Set: TOTAL

20 Linear Model: 485/589 (82.3%)

21 Non-Linear Model: 488/589 (82.9%)

22 Validation Set: Critically Evaluated as "Readily Degradable"

23 Italian (Italy) Linear Model: 105/131 (80.2%)

24 Non-Linear Model: 103/131 (78.6%)

25 Validation Set: Critically Evaluated as "Not Readily Degradable"

26 Italian (Italy) Linear Model: 135/164 (82.3%)

27 Non-Linear Model: 135/164 (82.3%)

28 Validation Set: TOTAL

29 Linear Model: 240/295 (81.3%)

30 Non-Linear Model: 238/295 (80.7%)
31
32

33 Accuracy of **HENRYWIN**: The accuracy of the bond contribution method is
34 discussed in detail in Meylan and Howard (1991). Briefly, a correlation
35 coefficient (R^2) of 0.97, a standard deviation (sd) of 0.34 and a mean error (me) of
36 0.21 were found for a 345 compound training set (all statistics apply to LWAPC
37 values). A 74 compound validation dataset had respective R^2 , sd and me statistics
38 of 0.96, 0.46 and 0.31. SRC's current experimental database contains 1650
39 compounds. Since publication of the Meylan and Howard (1991) article, the
40 methodology was updated (HENRYWIN version 2) by adding new bond
41 contribution values and new correction factors, especially for various classes of
42 pesticides.
43

44 At times, the bond estimate and the group estimate made by HENRYWIN may
45 vary significantly. Experience with HENRYWIN has shown that the difference
46 between bond and group methods can vary by as much as 2 orders of magnitude

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1 for some compounds with many functional groups. The estimation from the group
2 method is sometimes preferred unless the bond method uses a correction factor
3 from Table D-3 (Appendix D) or Appendix F. A recent independent evaluation
4 (Altschuh et al., 1999) for a diverse set of organic chemicals found the bond
5 method more accurate than the group method. The group method generates
6 inaccurate estimates for certain types of structures, such as
7 hexachlorocyclohexanes (Altschuh et al., 1999). At times, averaging two widely
8 divergent values is appropriate. For some compounds, both methods can yield a
9 Henry's Law constant of 1.0×10^{-12} atm-m³/mole or smaller. Numbers smaller than
10 this value may be unrealistically low. However, any organic compound with a
11 Henry's Law constant less than 3.0×10^{-7} is considered essentially non-volatile
12 from water (Thomas, 1990). The Exposure Evaluation Branch of the U.S. EPA
13 (OPPT) uses a cut-off of 1.0×10^{-8} atm- m³/mole for HLC estimates; any estimate
14 less than the cut-off is considered 1.0×10^{-18} atm- m³/mole.

15
16 Estimation Accuracy of **KOWWIN**: The figures in this Help file (not shown)
17 illustrate KOWWIN's ability to estimate accurate log P values. The listing
18 compares the accuracy of KOWWIN to the ClogPtm Program (Daylight, 1995;
19 BioByte, 1995) statistics using SRC's Experimental Log P Database: (n = number
20 of compounds; R = correlation coefficient; sd = standard deviation; me = absolute
21 mean error)

22 KOWWIN v1.63

23 Total: n=12805; R²=0.95; sd=0.435; me=0.316

24 Training: n=2474 R²=0.981 sd=0.22 me=0.16

25 Validation:n=10331 R²=0.94 sd=0.47 me=0.35

26
27 CLOGP for Windows (v1.0)

28 Total: n=11735(a) R²=0.91 sd=0.59 me=0.384

29 CLOGP (UNIX version as reported by Leo, 1992)

30 Total: n=7250 R²=0.96 sd=0.3

31 (using equation: Log P = 0.914 CLOGP + 0.184) (b)

32 ^(a) Taken from the current database; the difference between the entire
33 database (12686) and the number used (11616) is primarily due to
34 "missing fragments" in the CLOGP program. BioByte's Internet website
35 reports the following statistics for its starlist: n=8942, R²=0.917, sd=0.482
36 using the equation: Log P = 0.876CLOGP + 0.307.

37 ^(b) These statistics were determined after removing large systemic deviant
38 compounds and other large deviant structures where the underlying
39 difficulty is conformational (Leo, A.J. 1992. 30 years of calculating Log
40 P_{oct}. QSAR Meeting. July 23, 1992).

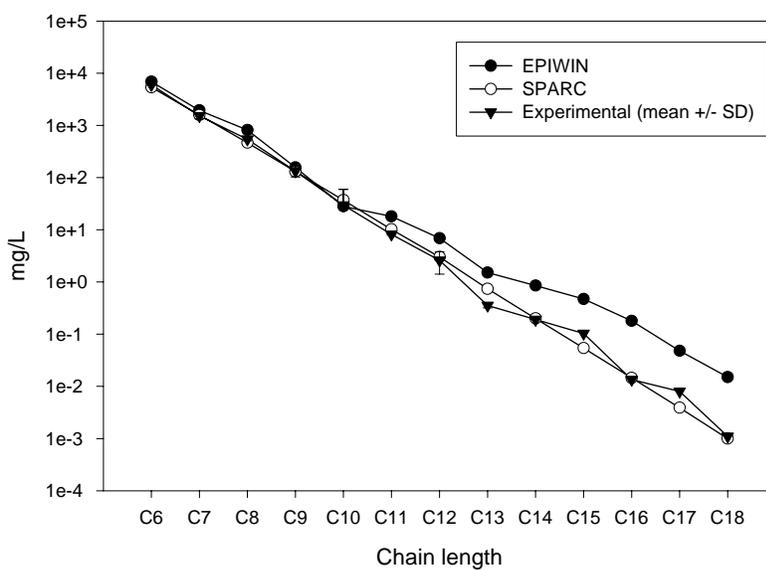
APPENDIX for 1Ci

The primary regulatory obligation in a tiered approach to risk assessment is conservatism of the prediction at the lowest tier (model based screening), not accuracy; tolerance towards false negatives differs between countries. The OECD HPV group is currently running an assessment of member countries' appreciation and application of (Q)SAR estimates. The results of this effort would lend itself useful to the EPA in reviewing the EPI SuiteTM. The EPA should consider all the listed criteria in the appendix when upgrades to the models are made.

Case study I: Water Solubility. Estimation of long chained aliphatic alcohols water solubility, comparative analysis between EPI SuiteTM (WSKOWWIN), SPARC, and measured

In relation to an HPV submission, a comparison of water solubility estimations for aliphatic alcohols, found that, for shorter-chain alcohols (C6-C10), the modeled and measured values were comparable. For mid-chain (C10-14) alcohols, the EPI SuiteTM model moderately overestimated the water solubility. For the longer-chain alcohols (C14-C18), the EPI SuiteTM overestimated water solubility by approximately one log unit, which could have an impact on the need for further toxicity assessment. This case illustrates that empirical regression driven models are more susceptible to error when screening complex compounds with few empirical data or data of questionable quality close to the limit of solubility, than thermal and quantum energy driven models such as SPARC, which are less dependent on measured values (Hilal, et al., 2003 a and b) see Figure in Appendix.

Long Alcohols Water Solubility



27
28

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1

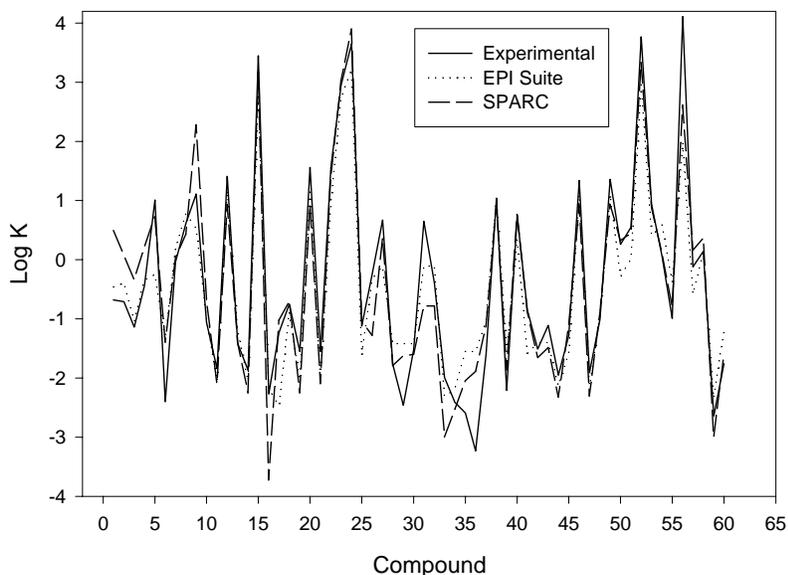
2 Case study II: Hydrolysis. Comparative analysis between EPI SuiteTM (HYDROWIN),
3 SPARC, and measured

4

5 For hydrolysis, SPARC, at this time, calculates only carboxylic acid ester hydrolysis rate
6 constants in any single or mixed solvent at any temperature. EPI SuiteTM calculates
7 esters, carbamates, epoxides, halomethanes, and alkyl chlorides hydrolysis rate constants
8 only in water. The SPARC residual mean squares deviation error of the calculated versus
9 observed values in water is better than 0.37 and R^2 equal to 0.98 while the EPI R^2 for 124
10 ester compounds is 0.965 (see appendix for list of compounds). Below are graphs
11 comparing the SPARC versus EPI SuiteTM calculations for carboxylic acid ester
12 hydrolysis rate. The end result for the 61 compounds is that SPARC does slightly better
13 in the mean unsigned error, but has a lower frequency of potential significant outliers
14 than EPI SuiteTM does – see graphs (Hilal, personal communication, 2005; Long Chained
15 Aliphatic Alcohols SIAR, 2006).

16

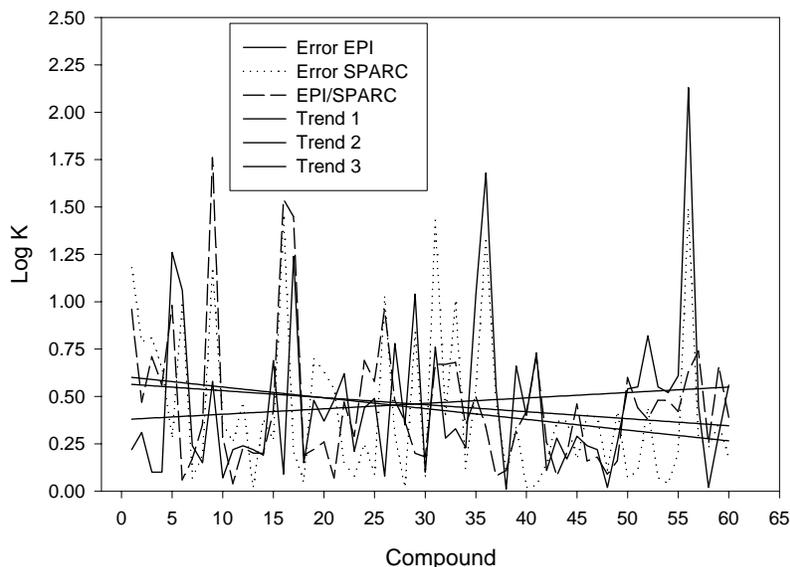
Hydrolysis Log K



17

18

Error (experimental +/- predicted)



1
2 Biodegradation (BIOWIN) and other fate models are more complicated than most of the
3 other algorithm driven (Q)SARs under EPI SuiteTM. Modeling of biodegradation has been
4 challenged due to lack of data and variability in soils (e.g. microorganism communities)
5 by IUPAC (Peijnenburg 1994). For K_{ow} (octanol/water) partition coefficient, EPI Suite
6 TM estimations are slightly better than SPARC especially when the log $K_{ow} < 7$ (which is
7 borderline for any model and experimental test). At higher K_{ow} , SPARC calculations are
8 better than EPI SuiteTM. At higher K_{ow} 's EPI SuiteTM is bound to measurements that
9 were later (through slow stir method) shown to be inaccurate. SPARC models K_{ow} as a
10 ratio of activity coefficient calculations. Originally, SPARC versus EPI Suite^M
11 calculations indicated that SPARC values are too high. After slow stir, SPARC calculated
12 the same K_{ow} value but experimental values changed and the SPARC values, though they
13 did not change, were now in better agreement with experimental values. For boiling
14 point, solubility and Henry's constant, SPARC performed well, and out-performed EPI
15 SuiteTM ($R^2 = 0.999$) (Hilal, et al. a and b). Total mean across all EPI SuiteTM R^2 best case
16 (using the highest R^2) = 0.95 ± 0.03 SD, and worst-case (lowest R^2) = 0.86 ± 0.15 SD.
17

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1

Table 1: Compounds in graphs in numerical order

Smiles (1-62)

O=C(OCC=C)C
c(ccc1COC(=O)C)cc1
O=C(OC(C)C=C)C
O=C(OC(C)C#C)C
O=C(OC=C)C
O=C(OC(C)(CC)C=C)C
O=C(OCC)CSC
O=C(OCC)CS(=O)C
O=C(OCC)CS(=O)(=O)C
O=C(OC)C
O=C(OC)C=CC
O=C(OCC)
O=C(OCC)CCC
O=C(OCC)C=CC
O=C(OC)C(Cl)Cl
O=C(OC)C=C(C)C
O=C(OC)C(C)=C
O=C(OC)C
O=C(OCC)C=CC
O=C(OC)
O=C(OC)C=CC
O=C(OCC)C(Cl)
O=C(OCC)C(Cl)Cl
O=C(OCC)C(F)F
O=C(OCC)C=C
O=C(OCC)C#CC
O=C(OCC)C#C
c12c(C(=O)OCC)cccc1cccc2
c12cc(C(=O)OCC)ccc1cccc2
O=C(OCC)c(cccc1)c1
O=C(OCC)C=CC(=O)OCC
O=C(OCC)C=CC(=O)OCC
O=C(OC(C)C(C))C=CC
O=C(OC(C)C)C=CC
c12c(C(=O)OC(C)C)cccc1cccc2
c12cc(C(=O)OC(C)C)ccc1cccc2
O=C(OC(C)C)C
O=C(OC(C)C)
O=C(OC(C)C)c(cccc1)c1
O=C(Oc(cc(N(=O)=O)c1)cc1)C=C

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O=C(OC)C=C
c12c(C(=O)OC)cccc1cccc2
c12cc(C(=O)OC)ccc1cccc2
O=C(OCCCC)C=CC
O=C(OCCCC)C=C
O=C(OCCCC)
O=C(OCCC)C=CC
O=C(OCCC)C
O=C(OCCC)
O=C(Oc(ccc(Cl)c1)c1)C=C
O=C(Oc(ccc(C(=O)C)c1)c1)C=C
O=C(Oc(ccc(N(=O)=O)c1)c1)C(Cl)
O=C(Oc(ccc(N(=O)=O)c1)c1)C=C
c12c(C(=O)(Oc(ccc(N(=O)=O)c3)c3))cccc1cccc2
c12c(C(=O)(Oc(ccc(OC)c3)c3))cccc1cccc2
O=C(Oc(cccc1)c1)C(Cl)
O=C(Oc(cccc1)c1)C=C
O=C(Oc(cccc1)c1)C
O=C(OC(C)CC)C=CC
O=C(OC(C)CC)C

1

GLOSSARY

1		
2		
3	AOPWIN	Atmospheric Oxidation Estimation Pestimation rogram
4	BAF	Bioaccumulation factor
5	BCF	Bioconcentration factor
6	BCFWIN	Bioconcentration factor estimation program
7	BIOWIN	Biodegradation factor estimation program
8	CAS Number	Chemical Abstract Services Registry Number
9	Chemdraw™	Chemical Drawing Program - CambridgeSoft Corporation
10	DERMWIN	Dermal Permeability Coefficient Program
11	DOS	Disk operating system
12	DSL	Domestic Substances List – Environment Canada
13	ECOSAR	Ecological Structure Activity Relationship Program
14	EFM	Environmental Fate Models
15	EPI	Estimation Program Interface
16	GAO	Government Accountability Office
17	GHS	Globally Harmonized System for Classification of Chemicals
18	GUI	Graphic User Interface
19	H _c	Henry's Law Constant
20	HENRYWIN	Henry's Law Constant Estimation Program
21	HPV	High Production Volume Chemicals
22	HYDROWIN	Hydrolysis Factor Estimation Program
23	IUPAC	International Union of Pure and Applied Chemistry
24	K _{AW}	Air-water Partitioning Coefficient
25	K _{OA}	Octanol/Octonal-Air Partitioning Coefficient
26	K _{OC}	Organic Carbon Partitioning Coefficient
27	K _{OW}	Octanol-Water Partitioning Coefficient
28	KOWWIN	Octanol-Water Partitioning Coefficient Estimation Program
29	LEVEL3NT	Level 3 Fugacity Estimation Program
30	MCI	Molecular Connectivity Indices
31	MDL	Elsevier Molecular Design Limited (MDL) Information
32		Systems
33	MTC	Mass Transfer Coefficient
34	MPBPWIN	Melting Point-Boiling Point Chemical Estimation Program
35	NAPL	Non-aqueous Phase liquid
36	OECD	Organization of Economic Cooperation and Development
37	OPP	Office of Pesticide Programs
38	OPPT	Office of Pollution Prevention and Toxics
39	OPPTS	Office of Prevention, Pesticides and Toxic Substances
40	OSWER	Office of Solid Waste and Emergency Response
41	PCKOCWIN	Organic Carbon Partitioning Coefficient Estimation Program
42	PER	Property Estimation Routine
43	pK _a	Negative Log of a Chemical's Dissociation Constant
44	PMN	Premanufacture Notice
45	POP	Persistent Organic Pollutants

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1	PP-LFER	Polyparameter Linear Free Energy Relationships
2	QSAR	Quantitative Structure Activity Relationship
3	QSPR	Quantitative Structure Property Relationship
4	REACH	European Union's Registration, Evaluation and Authorisation of Chemicals Policy
5		
6	SAICM	United Nations (UN) Strategic Approach to International Chemicals Management
7		
8	SPARC	Sparc Performs Automated Reasoning in Chemistry - http://ibmlc2.chem.uga.edu/sparc/
9		
10	SMILES	Simplified Molecular Input Line Entry System
11	SPC	Structure Property Correlation
12	STPWIN	Sewage Treatment Plant Chemical Fate Estimation Program
13	TSCA	Toxic Substances Control Act
14	UVCB	Unknown or Variable Composition, Complex Reaction Products and Biological Materials [per HS]
15		
16	WATERNT	Organic Compound Water Solubility Program
17	WSKOWIN	Water Solubility Estimation Program
18	WVOLVIN	Volatilization Rate from Water Estimation Program
19	VOC	Volatile Organic Compound
20		
21		